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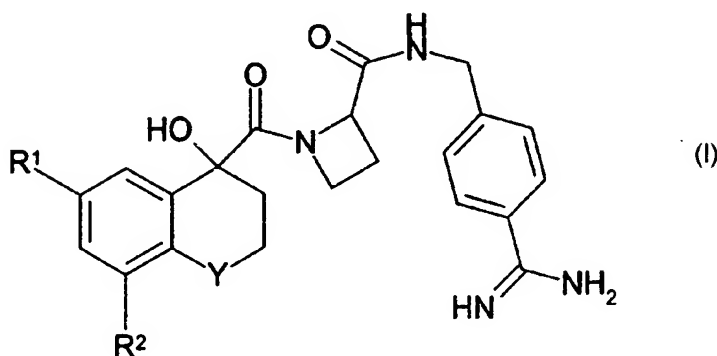
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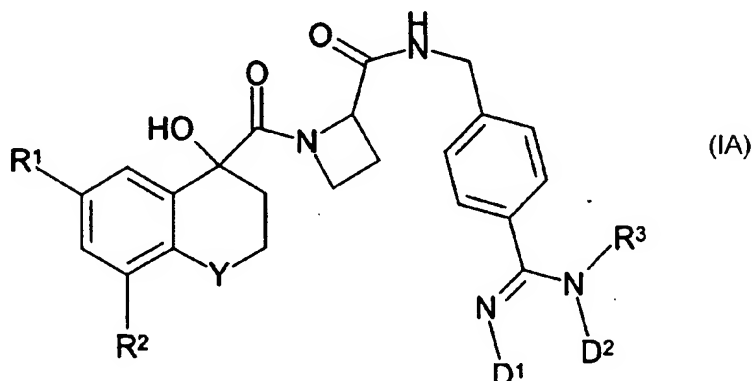
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**(54) Title:** NEW THIOCHROMANE DERIVATIVES AND THEIR USE AS THROMBIN INHIBITORS



**(57) Abstract:** There is provided compounds of formulae (I) and (IA) wherein Y, R<sup>1</sup>, R<sup>2</sup>, R<sup>3</sup>, D<sup>1</sup> and D<sup>2</sup> have meanings given in the description which are useful as, or as prodrugs of, competitive inhibitors of trypsinlike proteases, such as thrombin, and in particular in the treatment of conditions where inhibitors of thrombin is required (e.g. thrombosis) or as anticoagulants.





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*For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.*

## NEW THIOCHROMANE DERIVATIVES AND THEIR USE AS THROMBIN INHIBITORS

### Field of the Invention

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This invention relates to novel pharmaceutically useful compounds, in particular compounds that are, or are prodrugs of, competitive inhibitors of trypsin-like serine proteases, especially thrombin, their use as medicaments, pharmaceutical compositions containing them and synthetic  
10 routes to their production.

### Background

Blood coagulation is the key process involved in both haemostasis (i.e. the  
15 prevention of blood loss from a damaged vessel) and thrombosis (i.e. the formation of a blood clot in a blood vessel, sometimes leading to vessel obstruction).

Coagulation is the result of a complex series of enzymatic reactions. One  
20 of the ultimate steps in this series of reactions is the conversion of the proenzyme prothrombin to the active enzyme thrombin.

Thrombin is known to play a central role in coagulation. It activates platelets, leading to platelet aggregation, converts fibrinogen into fibrin  
25 monomers, which polymerise spontaneously into fibrin polymers, and activates factor XIII, which in turn crosslinks the polymers to form insoluble fibrin. Furthermore, thrombin activates factor V and factor VIII leading to a "positive feedback" generation of thrombin from prothrombin.

## 2.

By inhibiting the aggregation of platelets and the formation and crosslinking of fibrin, effective inhibitors of thrombin would be expected to exhibit antithrombotic activity. In addition, antithrombotic activity would be expected to be enhanced by effective inhibition of the positive  
5 feedback mechanism.

Further, it is known that administration of prodrugs of thrombin inhibitors may give rise to improvements in:

- (a) certain pharmacokinetic properties after administration of; and
- 10 (b) the prevalence of certain side effects associated with, those inhibitors.

**Prior Art**

- 15 The early development of low molecular weight inhibitors of thrombin has been described by Claesson in Blood Coagul. Fibrinol. (1994) 5, 411.

Blombäck *et al* (in J. Clin. Lab. Invest. 24, suppl. 107, 59, (1969)) reported thrombin inhibitors based on the amino acid sequence situated  
20 around the cleavage site for the fibrinogen A $\alpha$  chain. Of the amino acid sequences discussed, these authors suggested the tripeptide sequence Phe-Val-Arg (P9-P2-P1, hereinafter referred to as the P3-P2-P1 sequence) would be the most effective inhibitor.

- 25 Thrombin inhibitors based on dipeptidyl derivatives with an  $\alpha,\omega$ -aminoalkyl guanidine in the P1-position are known from US Patent N $^{\circ}$  4,346,078 and International Patent Application WO 93/11152. Similar, structurally related, dipeptidyl derivatives have also been reported. For example International Patent Application WO 94/29336 discloses

compounds with, for example, aminomethyl benzamidines, cyclic aminoalkyl amidines and cyclic aminoalkyl guanidines in the P1-position (International Patent Application WO 97/23499 discloses prodrugs of certain of these compounds); European Patent Application 0 648 780,  
5 discloses compounds with, for example, cyclic aminoalkyl guanidines in the P1-position.

Thrombin inhibitors based on peptidyl derivatives, also having cyclic aminoalkyl guanidines (e.g. either 3- or 4- aminomethyl-1-amidino-  
10 piperidine) in the P1-position are known from European Patent Applications 0 468 231, 0 559 046 and 0 641 779.

Thrombin inhibitors based on tripeptidyl derivatives with arginine aldehyde in the P1-position were first disclosed in European Patent  
15 Application 0 185 390.

More recently, arginine aldehyde-based peptidyl derivatives, modified in the P3-position, have been reported. For example, International Patent Application WO 93/18060 discloses hydroxy acids, European Patent  
20 Application 0 526 877 des-amino acids, and European Patent Application 0 542 525 O-methyl mandelic acids in the P3-position.

Inhibitors of serine proteases (e.g. thrombin) based on electrophilic ketones in the P1-position are also known. For example, European Patent  
25 Application 0 195 212 discloses peptidyl  $\alpha$ -keto esters and amides, European Patent Application 0 362 002 fluoroalkylamide ketones, European Patent Application 0 364 344  $\alpha,\beta,\delta$ -triketocompounds, and European Patent Application 0 530 167  $\alpha$ -alkoxy ketone derivatives of arginine in the P1-position.

Other, structurally different, inhibitors of trypsin-like serine proteases based on C-terminal boronic acid derivatives of arginine and isothiuronium analogues thereof are known from European Patent Application 0 293 881.

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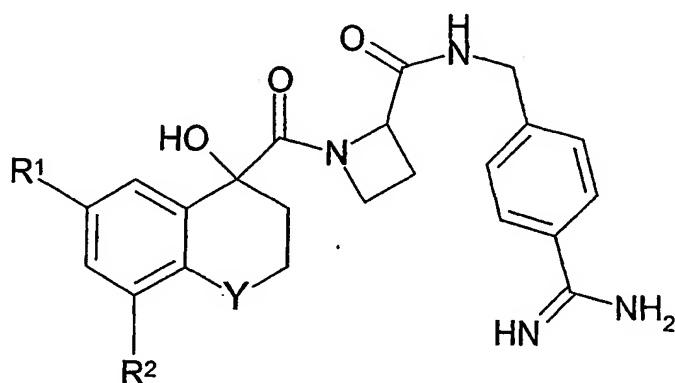
More recently, thrombin inhibitors based on peptidyl derivatives have been disclosed in European Patent Application 0 669 317 and International Patent Applications WO 95/35309, WO 95/23609, WO 96/25426, WO 97/02284, WO 97/46577, WO 96/32110, WO 96/31504, WO 96/03374, 10 WO 98/06740, WO 97/49404 and WO 99/29664. Certain prodrugs of thrombin inhibitors have been disclosed in WO 97/33576.

WO 98/57932 and WO 00/35869 disclose thrombin inhibitors, and prodrugs of thrombin inhibitors, based on peptidyl derivatives with fused, 15 bi- or tri-cyclic acids at the P3-position.

However, there remains a need for effective inhibitors of trypsin-like serine proteases, such as thrombin. There is also a need for compounds that have a favourable pharmacokinetic profile (e.g. low clearance), are 20 orally bioavailable, and are selective in inhibiting thrombin over other serine proteases, in particular those involved in haemostasis. Compounds which exhibit competitive inhibitory activity towards thrombin would be expected to be especially useful as anticoagulants and therefore in the therapeutic treatment of thrombosis and related disorders.

**Disclosure of the Invention**

According to the invention there is provided compounds of formula I,



5

wherein

Y represents S(O) or S(O)<sub>2</sub>;

R<sup>1</sup> represents halo; and

R<sup>2</sup> represents H, halo or C<sub>1-4</sub> alkoxy (which latter group is optionally substituted by one or more halo groups);

10

or a pharmaceutically acceptable derivative thereof,

which compounds are referred to hereinafter as “the compounds of the invention”.

15 The term “pharmaceutically acceptable derivatives” of compounds of formula I includes pharmaceutically acceptable salts. Suitable salts include inorganic acid (e.g. hydrogen halide), and organic acid (e.g. acetic, methanesulfonic or trifluoroacetic acid), addition salts.

20 The alkyl part of alkoxy groups which R<sup>2</sup> may represent, may, when there is a sufficient number of carbon atoms, be linear or branched, be saturated

or unsaturated, be cyclic, acyclic or part cyclic/acyclic, and/or be optionally interrupted by an O atom.

Halo groups which  $R^1$  and  $R^2$  may represent, and with which  $R^2$  may be substituted, include fluoro, chloro, bromo and iodo.

Abbreviations are listed at the end of this specification.

Compounds of the invention that may be mentioned include those in which Y represents  $S(O)_2$ .

Preferred compounds of the invention include those in which:

$R^1$  represents chloro;

$R^2$  represents H, halo or  $C_{1-2}$  alkoxy (which latter group is optionally substituted by one or more halo (e.g. fluoro) groups).

More preferred compounds of the invention include those in which:

$R^1$  represents chloro; and

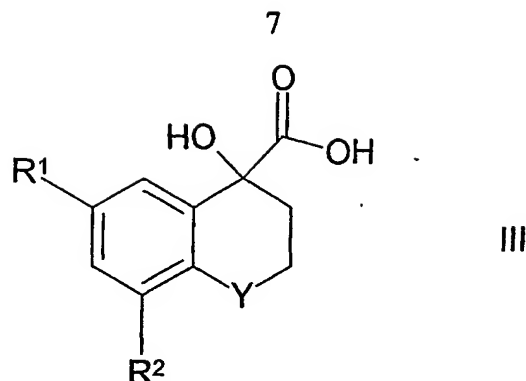
$R^2$  represents H, chloro,  $OCHF_2$ ,  $OCF_3$  or, especially,  $OCH_3$ .

Preferred compounds of formula I include the compounds of the Examples described hereinafter, particularly the compound of Example 1.

According to the invention there is also provided a process for the preparation of compounds of formula I which comprises:

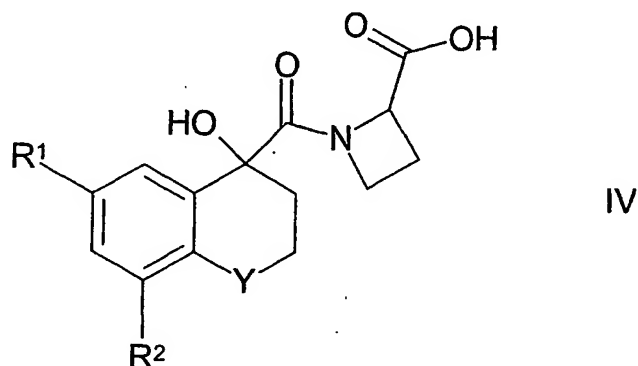
(i) the coupling of a compound of formula III,





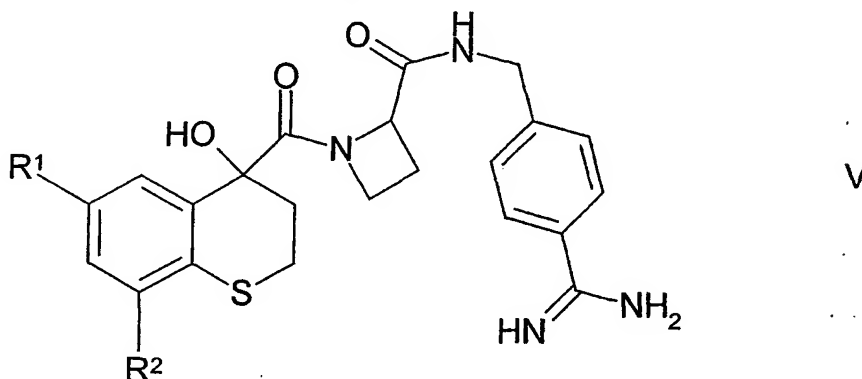
wherein Y, R<sup>1</sup> and R<sup>2</sup> are as hereinbefore defined, with 4-amidinobenzyl-2-azetidinecarboxamide (see, for example, international patent application WO 97/02284), for example in the presence of a coupling agent (e.g. EDC, DCC, HBTU, HATU, TBTU, PyBOP or oxalyl chloride in DMF), an appropriate base (e.g. pyridine, 2,4,6-collidine, DMAP, TEA or DIPEA) and a suitable organic solvent (e.g. dichloromethane, acetonitrile or DMF);

(ii) the coupling of a compound of formula IV,



wherein Y, R<sup>1</sup> and R<sup>2</sup> are as hereinbefore defined, with *para*-amidino-benzylamine, for example in the presence of a coupling agent (e.g. oxalyl chloride in DMF, EDC, DCC, HBTU, HATU, PyBOP or TBTU), an appropriate base (e.g. pyridine, DMAP, TEA, 2,4,6-collidine or DIPEA) and a suitable organic solvent (e.g. dichloromethane, acetonitrile or DMF); or

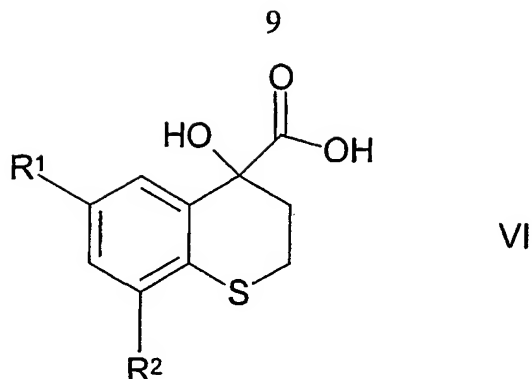
(iii) complete oxidation (for compounds of formula I in which Y is S(O)<sub>2</sub>), or partial oxidation (for compounds of formula I in which Y is S(O)), of a corresponding compound of formula V,



5 wherein R<sup>1</sup> and R<sup>2</sup> are as hereinbefore defined, for example in the presence of an appropriate amount of a suitable oxidising agent (e.g. *m*CPBA, hydrogen peroxide or potassium peroxymonosulfate) and an appropriate organic solvent (e.g. CH<sub>2</sub>Cl<sub>2</sub>, methanol, water or mixtures thereof), and  
10 optionally in the presence of a suitable protic acid (e.g. acetic acid).

The skilled person will appreciate that, in the case of partial oxidation, a mixture of stereoisomers may be obtained, which may be separated by techniques known to those skilled in the art (e.g. by column  
15 chromatography or chiral chromatography).

Compounds of formula III may be prepared by complete or partial oxidation of a compound of formula VI,



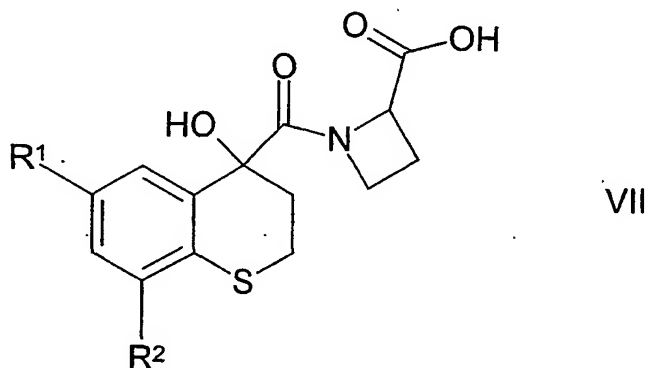
wherein  $R^1$  and  $R^2$  are as hereinbefore defined, for example under conditions such as those described hereinbefore for the synthesis of compounds of formula I (process step (iii)).

5

Compounds of formula IV may be prepared by the coupling of a compound of formula III, as hereinbefore defined, with azetidine-2-carboxylic acid, for example under conditions such as those described hereinbefore for synthesis of compounds of formula I (see, for example,

10 process steps (i) and (ii)).

Compounds of formula IV may alternatively be prepared by complete or partial oxidation of a compound of formula VII,

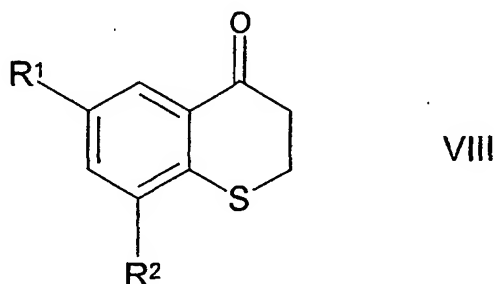


15 wherein  $R^1$  and  $R^2$  are as hereinbefore defined, for example under conditions such as those described hereinbefore for the synthesis of compounds of formula I (process step (iii)).

Compounds of formula V may be prepared in accordance with peptide coupling techniques, for example in analogous fashion to the methods described hereinbefore for compounds of formula I (see, for example, process steps (i) and (ii)). If desired, compounds of formula VII may also be prepared in this way.

Compounds of formula VI are available using known and/or standard techniques.

For example, compounds of formula VI may be prepared by reaction of a compound of formula VIII,



wherein  $R^1$  and  $R^2$  are as hereinbefore defined, with:

(a) a compound of formula IX,



wherein  $R''$  represents H or  $(CH_3)_3Si$ , for example at room, or elevated, temperature (e.g. below  $100^\circ C$ ) in the presence of a suitable organic solvent (e.g. chloroform or dichloromethane) and, if necessary, in the presence of a suitable base (e.g. TEA) and/or a suitable catalyst system (e.g. benzylammonium chloride or zinc iodide), followed by hydrolysis in the presence of an acid (e.g. HCl or  $H_2SO_4$ ), for example at  $20^\circ C$  (e.g. according, or analogously, to the method described by C.F. Bigge *et al.* in

J. Med. Chem. (1993) 36, 1977), and then followed by hydrolysis under alkaline conditions (e.g. in the presence of water and either lithium or potassium hydroxide) to give the free acid;

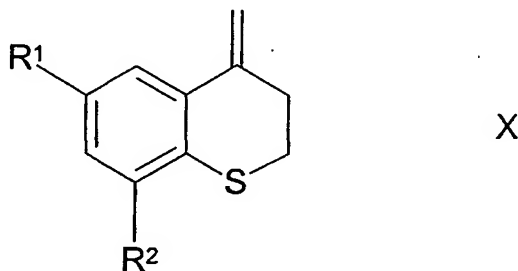
- 5 (b) NaCN or KCN, for example in the presence of NaHSO<sub>3</sub> and water, followed by hydrolysis; or
- (c) chloroform, for example at elevated temperature (e.g. above room temperature but below 100°C) in the presence of a suitable base (e.g. sodium hydroxide) and, if necessary, in the presence of a suitable catalyst
- 10 system (e.g. benzylammonium chloride), followed by hydrolysis.

The enantiomeric forms of compounds of formula VI (i.e. those compounds having different configurations of substituents about the

15 C-atom that is in the  $\alpha$ -position relative to the CO<sub>2</sub>H group) may be separated by techniques known to those skilled in the art (e.g. by chromatography, using a chiral chromatographic medium).

Compounds of formula VI may alternatively be prepared by way of a

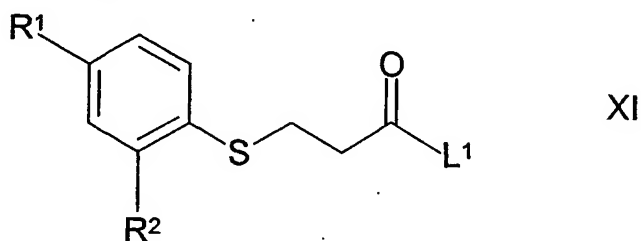
20 Sharpless stereoselective dihydroxylation of a compound of formula X,



wherein R<sup>1</sup> and R<sup>2</sup> are as hereinbefore defined, under conditions known to those skilled in the art (e.g. at low temperature (e.g. 0°C), using, for example, the commercial reagent AD-mix- $\beta^{\text{TM}}$  in the presence of suitable

solvent (e.g. *t*-butanol), followed by oxidation of the resultant intermediate (e.g. at elevated temperature (e.g. 75°C) in the presence of a stream of air and Pt/C (5%) in acetone/water).

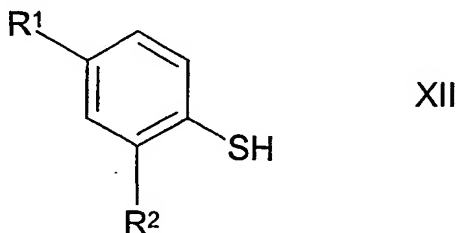
- 5 Compounds of formula VIII are available using known and/or standard techniques. For example, compounds of formula VIII may be prepared by cyclisation of a compound of formula XI,



- wherein L<sup>1</sup> represents a leaving group such as OH, C<sub>1-4</sub> alkoxy, C<sub>1-4</sub> acyloxy (optionally substituted by one or more halo atoms) or halo and R<sup>1</sup> and R<sup>2</sup> are as hereinbefore defined, for example in the presence of a suitable catalyst (e.g. trifluoroacetic anhydride, a protic acid such as H<sub>2</sub>SO<sub>4</sub> or a Lewis acid such as BF<sub>3</sub>) and optionally in the presence of an appropriate solvent (e.g. CH<sub>2</sub>Cl<sub>2</sub>).

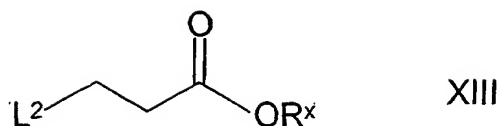
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Compounds of formula XI may be prepared in accordance with known techniques. For example, compounds of formula XI may be prepared by reaction of a compound of formula XII,



- 20 wherein R<sup>1</sup> and R<sup>2</sup> are as hereinbefore defined, with a compound of formula XIII,

13



- wherein  $L^2$  represents a leaving group such as halo and  $R^x$  represents  $C_{1-6}$  alkyl, for example at between room and reflux temperature in the presence of an appropriate base (e.g. triethylamine or  $Cs_2CO_3$ ) and a suitable solvent (e.g. ethyl acetate or acetone), followed by conversion of the  $OR^x$  group into an  $L^1$  group under conditions that are well known to those skilled in the art (e.g. for compounds of formula XI in which  $L^1$  represents OH, by hydrolysis under alkaline conditions).
- 10 Compounds of formulae IX, X, XII and XIII, and derivatives thereof, are either commercially available, are known in the literature, or may be obtained either by analogy with the processes described herein, or by conventional synthetic procedures, in accordance with standard techniques, from readily available starting materials using appropriate reagents and reaction conditions (e.g. as described hereinafter).
- 15

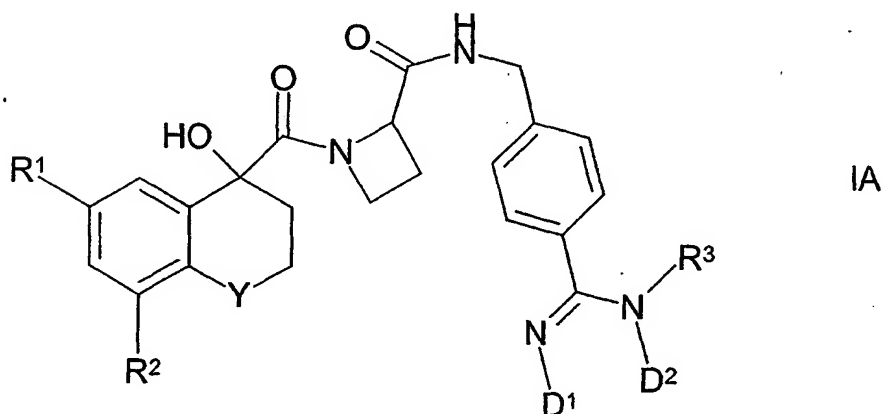
The compounds of formula I may be isolated from their reaction mixtures using conventional techniques.

- 20 Substituents on the aromatic ring in compounds of formulae I, III, IV, V, VI, VII, VIII, X, XI and XII may be introduced and/or interconverted using techniques well known to those skilled in the art. For example, halo may be introduced, both into the aromatic ring and into the alkyl part of alkoxy groups that  $R^2$  may represent, by reaction with suitable halogenating agents, for example as described hereinafter.
- 25

In accordance with the present invention, pharmaceutically acceptable derivatives of compounds of formula I also include "protected" derivatives of compounds of formula I and/or compounds that act as prodrugs of compounds of formula I.

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In this respect, according to a further aspect of the invention there is provided derivatives of compounds of formula I as defined herein which are compounds of formula IA,



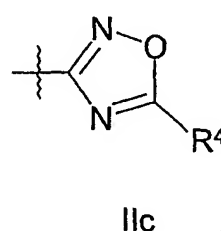
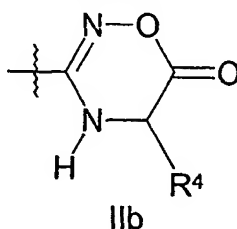
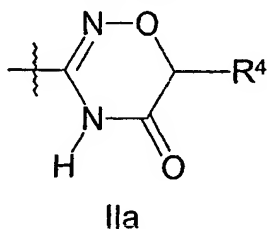
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wherein

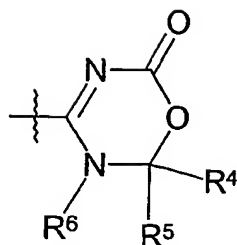
Y, R<sup>1</sup> and R<sup>2</sup> are as hereinbefore defined;

D<sup>1</sup> and D<sup>2</sup> independently represent H, -OR<sup>7</sup> or R<sup>8</sup>, or D<sup>1</sup>, D<sup>2</sup> and R<sup>3</sup>, together with the amidine group to which they are attached, form a cyclic

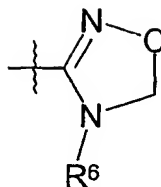
15 group of formula IIa, IIb, IIc, IId or Ile,







IId



IIe

wherein wavy lines indicate the points of attachment to the benzene ring;

R<sup>3</sup> represents H, or R<sup>3</sup>, D<sup>1</sup>, and D<sup>2</sup> together with the amidine group to which they are attached, form a cyclic group of formula IIa, IIb, IIc, IId

5 or IIe;

R<sup>4</sup> and R<sup>5</sup> independently represent H or C<sub>1-4</sub> alkyl;

R<sup>6</sup> represents H, C<sub>1-6</sub> alkyl (which latter group is optionally substituted by one or more halo groups) or C(O)OR<sup>12</sup>;

R<sup>7</sup> represents H, C<sub>6-10</sub> aryl, C<sub>1-10</sub> alkyl (which latter group is optionally substituted by one or more halo groups), C<sub>1-3</sub> alkylphenyl, -C(R<sup>9a</sup>)(R<sup>9b</sup>)R<sup>10</sup>,  
10 -C(O)R<sup>11a</sup>, -C(O)OR<sup>12</sup>, -C(O)N(R<sup>13</sup>)R<sup>14</sup> or -(CH<sub>2</sub>)<sub>n</sub>(O)<sub>m</sub>R<sup>15</sup>;

R<sup>8</sup> represents -C(R<sup>9a</sup>)(R<sup>9b</sup>)R<sup>10</sup>, -C(O)R<sup>11b</sup> or -C(O)OR<sup>12</sup>;

R<sup>9a</sup> and R<sup>9b</sup> independently represent, at each occurrence, H or C<sub>1-6</sub> alkyl;

R<sup>10</sup> represents, at each occurrence, -OC(O)R<sup>16a</sup>, -OC(O)OR<sup>17</sup>,  
15 -N(R<sup>18a</sup>)C(O)OR<sup>17</sup> or -OC(O)N(R<sup>18b</sup>)R<sup>17</sup>;

R<sup>11a</sup> and R<sup>11b</sup> independently represent, at each occurrence, C<sub>6-10</sub> aryl, C<sub>1-3</sub> alkylphenyl (which latter two groups are optionally substituted by one or more substituents selected from C<sub>1-6</sub> alkyl and halo),  
-[C(R<sup>19a</sup>)(R<sup>19b</sup>)]<sub>p</sub>OC(O)R<sup>16b</sup>, or R<sup>11a</sup> represents C<sub>1-17</sub> alkyl (optionally substituted by C<sub>1-6</sub> alkoxy, C<sub>1-6</sub> acyloxy, amino or halo) or R<sup>11b</sup> represents  
20 C<sub>1-6</sub> alkyl;

R<sup>12</sup> represents, at each occurrence, C<sub>1-17</sub> alkyl (optionally substituted by one or more substituents selected from C<sub>1-6</sub> alkoxy, C<sub>1-6</sub> acyloxy, -Si(R<sup>20a</sup>)(R<sup>20b</sup>)(R<sup>20c</sup>) and halo), C<sub>6-10</sub> aryl, C<sub>1-3</sub> alkylphenyl (which latter two

groups are optionally substituted by  $C_{1-6}$  alkyl,  $C_{1-6}$  alkoxy, cyano or halo),  $-[C(R^{19a})(R^{19b})]_qOC(O)R^{16b}$  or  $-CH_2R^{21}$ ;

$R^{13}$  represents H or  $C_{1-7}$  alkyl, or together with  $R^{14}$  represents  $C_{4-5}$  alkylene;

- 5  $R^{14}$  represents  $C_{6-10}$  aryl or  $C_{1-10}$  alkyl (which latter group is optionally substituted by one or more substituents selected from OH, halo,  $CO_2H$ ,  $C_{1-6}$  alkoxy,  $C_{1-6}$  acyloxy and  $C_{6-10}$  aryl), or together with  $R^{13}$  represents  $C_{4-5}$  alkylene;

- $R^{15}$  represents  $C_{1-7}$  alkyl optionally substituted by one or more  
10  $-OC(O)C(H)(R^{22})N(G)(G^a)$  groups;

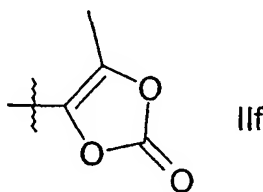
- $R^{16a}$ ,  $R^{16b}$  and  $R^{17}$  independently represent, at each occurrence,  $C_{6-10}$  aryl or  $C_{1-17}$  alkyl (which latter group is optionally substituted by one or more substituents selected from  $-OH$ , halo,  $-CO_2H$ ,  $C_{1-6}$  alkoxy,  $C_{1-6}$  acyloxy and  $C_{6-10}$  aryl), or  $R^{16b}$  represents  $C_{1-6}$  alkoxy (optionally substituted by  
15 one or more substituents selected from  $C_{1-6}$  alkyl and halo);

$R^{18a}$  and  $R^{18b}$  independently represent H or  $C_{1-4}$  alkyl;

$R^{19a}$  and  $R^{19b}$  independently represent, at each occurrence, H or  $C_{1-6}$  alkyl;

$R^{20a}$  to  $R^{20c}$  independently represent, at each occurrence,  $C_{1-6}$  alkyl or phenyl;

- 20  $R^{21}$  represents the structural fragment II f



$R^{22}$  represents  $C_{3-4}$  alkyl;

G and  $G^a$  independently represent H, an amino protective group, or G and  $G^a$  together represent an amino protective group;

- 25 m represents 0 or 1;

n represents 1, 2 or 3;

p represents 3 or 4;  
q represents 2 or 3;  
or a pharmaceutically acceptable salt thereof,

5 provided that:

- (a)  $D^1$  and  $D^2$  do not both represent H; and
- (b) when one of  $D^1$  and  $D^2$  represents  $-OR^7$ , then the other represents H.

Suitable "pharmaceutically acceptable salts" of compounds of formula IA  
10 include inorganic acid (e.g. hydrogen halide), and organic acid (e.g. acetic, methanesulfonic or trifluoroacetic acid), addition salts.

The compounds of the invention may exhibit tautomerism. All tautomeric forms and mixtures thereof are included within the scope of the invention.

15 Particular tautomeric forms that may be mentioned include those connected with the position of the double bond in the amidine functionality, and the position of the substituent  $D^1$  or  $D^2$ , in compounds of formula IA.

20 For the avoidance of doubt, in compounds of formula IA, substituents  $D^1$  and  $D^2$  are completely independent of each other. For example,  $D^1$  and  $D^2$  may both be represented by  $R^8$  in which, in both cases,  $R^8$  represents  $-C(O)R^{11b}$  in which  $R^{11b}$  represents, in both cases,  $C_{1-6}$  alkyl. However, in such instances the  $C_{1-6}$  alkyl group may be the same or different.

25

The compounds of formulae I and IA also contain at least two asymmetric carbon atoms and may therefore exhibit optical and/or diastereoisomerism. All diastereoisomers may be separated using conventional techniques, e.g. chromatography or fractional crystallisation. The various stereoisomers

may be isolated by separation of a racemic or other mixture of the compounds using conventional, e.g. fractional crystallisation or HPLC. techniques. Alternatively the desired optical isomers may be made by reaction of the appropriate optically active starting materials under  
5 conditions which will not cause racemisation or epimerisation, or by derivatisation, for example with a homochiral acid followed by separation of the diastereomeric derivatives by conventional means (e.g. HPLC. chromatography over silica). All stereoisomers are included within the scope of the invention.

10

As used herein, the term "aryl" includes phenyl, naphthyl (e.g. 2-naphthyl) and the like. Unless otherwise indicated, aryl groups are optionally substituted by one or more substituents selected from C<sub>1-6</sub> alkyl and halo.

15

Alkyl and alkylene groups, as well as alkyl parts of alkoxy, alkylphenyl and acyloxy groups, in compounds of formula IA may, when there is a sufficient number of carbon atoms, be linear or branched, be saturated or unsaturated, be cyclic, acyclic or part cyclic/acyclic, and/or be optionally  
20 interrupted by an O atom. The skilled person will appreciate that, when alkyl groups in compounds of formula IA are cyclic and interrupted by oxygen, they may then represent oxygen-containing heterocycles such as tetrahydrofuran-1-yl or (where appropriate) tetrahydropyranyl.

25 Halo groups in compounds of formula IA include fluoro, chloro, bromo and iodo.

As used herein, the term "amino protective group" includes groups mentioned in "Protective Groups in Organic Synthesis", 2<sup>nd</sup> edition, T W

Greene & P G M Wutz, Wiley-Interscience (1991), in particular those indexed at the start of the chapter entitled "*Protection for the Amino Group*" (see pages 309 to 315) of that reference, the disclosure in which document is hereby incorporated by reference.

5

Specific examples of amino protective groups thus include:

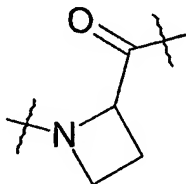
- (a) carbamate groups (e.g. methyl, cyclopropylmethyl, 1-methyl-1-cyclopropylmethyl, diisopropylmethyl, 9-fluorenylmethyl, 9-(2-sulfo)fluorenylmethyl, 2-furanylmethyl, 2,2,2-trichloroethyl, 2-haloethyl, 2-trimethylsilylethyl, 2-methylthioethyl, 2-methylsulfonylethyl, 2(*p*-toluenesulfonyl)ethyl, 2-phosphonioethyl, 1,1-dimethylpropynyl, 1,1-dimethyl-3-(*N,N*-dimethylcarboxamido)propyl, 1,1-dimethyl-3-(*N,N*-diethylamino)propyl, 1-methyl-1-(1-adamantyl)ethyl, 1-methyl-1-phenylethyl, 1-methyl-1-(3,5-dimethoxyphenyl)ethyl, 1-methyl-1-(4-biphenyl)ethyl, 1-methyl-1-(*p*-phenylazophenyl)ethyl, 1,1-dimethyl-2-haloethyl, 1,1-dimethyl-2,2,2-trichloroethyl, 1,1-dimethyl-2-cyanoethyl, isobutyl, *t*-butyl, *t*-amyl, cyclobutyl, 1-methylcyclobutyl, cyclopentyl, cyclohexyl, 1-methylcyclohexyl, 1-adamantyl, isobornyl, vinyl, allyl, cinnamyl, phenyl, 2,4,6-tri-*t*-butylphenyl, *m*-nitrophenyl, *S*-phenyl, 8-quinolinyl, *N*-hydroxypiperidinyl, 4-(1,4-dimethylpiperidinyl), 4,5-diphenyl-3-oxazolin-2-one, benzyl, 2,4,6-trimethylbenzyl, *p*-methoxybenzyl, 3,5-dimethoxybenzyl, *p*-decyloxybenzyl, *p*-nitrobenzyl, *o*-nitrobenzyl, 3,4-dimethoxy-6-nitrobenzyl, *p*-bromobenzyl, chlorobenzyl, 2,4-dichloro-benzyl, *p*-cyanobenzyl, *o*-(*N,N*-dimethylcarboxamidobenzyl)benzyl, *m*-chloro-*p*-acyloxybenzyl, *p*-(dihydroxyboryl)benzyl, *p*-(phenylazo)benzyl, *p*-(*p*'-methoxyphenylazo)benzyl, 5-benzisoxazolylmethyl, 9-anthrylmethyl, diphenylmethyl, phenyl(*o*-
- 10  
15  
20  
25

- nitrophenyl)methyl, di(2-pyridyl)methyl, 1-methyl-1-(4-pyridyl)-ethyl, isonicotinyl or *S*-benzyl carbamate groups);
- (b) amide groups (e.g. *N*-formyl, *N*-acetyl, *N*-chloroacetyl, *N*-dichloroacetyl, *N*-trichloroacetyl, *N*-trifluoroacetyl, *N*-*o*-nitrophenylacetyl, *N*-*o*-nitrophenoxyacetyl, *N*-acetoacetyl, *N*-acetylpyridinium, *N*-3-phenylpropionyl, *N*-3-(*p*-hydroxyphenyl)propionyl, *N*-3-(*o*-nitrophenyl)propionyl, *N*-2-methyl-2-(*o*-nitrophenoxy)propionyl, *N*-2-methyl-2-(*o*-phenylazophenoxy)propionyl, *N*-4-chlorobutyryl, *N*-isobutyryl, *N*-*o*-nitrocinnamoyl, *N*-picolinoyl, *N*-(*N*'-acetyl-methionyl), *N*-(*N*'-benzoylphenylalanyl), *N*-benzoyl, *N*-*p*-phenylbenzoyl, *N*-*p*-methoxybenzoyl, *N*-*o*-nitrobenzoyl or *N*-*o*-(benzoyloxy-methyl)benzoyl amide groups);
- (c) alkyl groups (e.g. *N*-allyl, *N*-phenacyl, *N*-3-acetoxypropyl, *N*-(4-nitro-1-cyclohexyl-2-oxo-pyrrolin-3-yl), *N*-methoxymethyl, *N*-chloroethoxymethyl, *N*-benzyloxymethyl, *N*-pivaloyloxymethyl, *N*-2-tetrahydropyranyl, *N*-2,4-dinitrophenyl, *N*-benzyl, *N*-3,4-dimethoxybenzyl, *N*-*o*-nitrobenzyl, *N*-di(*p*-methoxyphenyl)methyl, *N*-triphenylmethyl, *N*-(*p*-methoxyphenyl)diphenylmethyl, *N*-diphenyl-4-pyridylmethyl, *N*-2-picolyl *N*'-oxide or *N*-dibenzosuberyl groups);
- (d) phosphinyl and phosphoryl groups (e.g. *N*-diphenylphosphinyl, *N*-dimethylthiophosphinyl, *N*-diphenylthiophosphinyl, *N*-diethylphosphoryl, *N*-dibenzylphosphoryl or *N*-phenylphosphoryl groups);
- (e) sulfenyl groups (e.g. *N*-benzenesulfenyl, *N*-*o*-nitrobenzenesulfenyl, *N*-2,4-dinitrobenzenesulfenyl, *N*-pentachlorobenzenesulfenyl, *N*-2-nitro-4-methoxybenzenesulfenyl or *N*-triphenylmethylsulfenyl groups);
- (f) sulfonyl groups (e.g. *N*-benzenesulfonyl, *N*-*p*-methoxybenzenesulfonyl, *N*-2,4,6-trimethylbenzenesulfonyl, *N*-toluenesulfonyl, *N*-

benzylsulfonyl, *N*-*p*-methylbenzylsulfonyl, *N*-trifluoromethylsulfonyl or *N*-phenacylsulfonyl); and

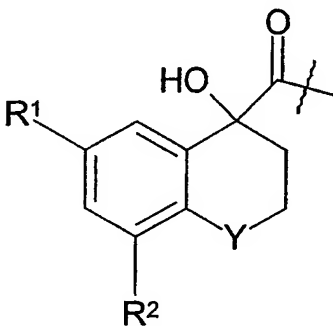
(g) the *N*-trimethylsilyl group.

5 Compounds of formulae I and IA in which the fragment



is in the *S*-configuration are preferred.

Compounds of formulae I and IA in which the fragment



10

is in the *R*-configuration are preferred.

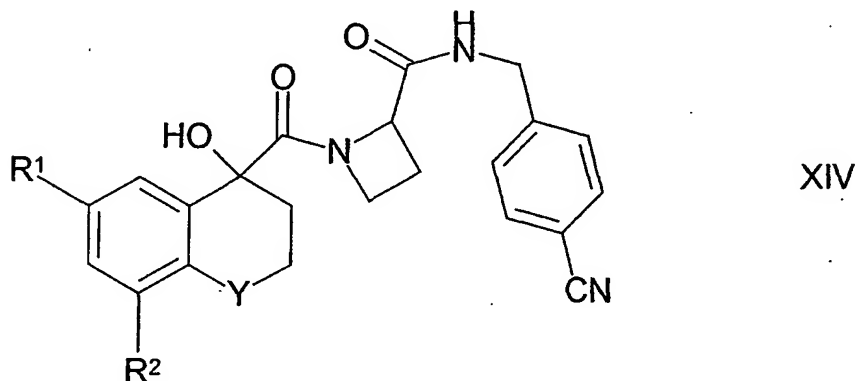
The wavy lines on the bonds in the above two fragments signify the bond positions of the fragments.

15

Compounds of formula IA may be prepared directly from compounds of formula I, in an analogous manner to compounds of formula I, or using procedures appropriately adapted from those used to prepare compounds of formula I. In this respect, processes that may be used to prepare  
20 compounds of formula IA include:

(1) for compounds of formula IA in which one of D<sup>1</sup> and D<sup>2</sup> represents -OR<sup>7</sup> (in which R<sup>7</sup> represents H, C<sub>6-10</sub> aryl, C<sub>1-10</sub> alkyl (which latter group is optionally substituted by one or more halo groups) or C<sub>1-3</sub> alkylphenyl), reaction of a compound of formula XIV,

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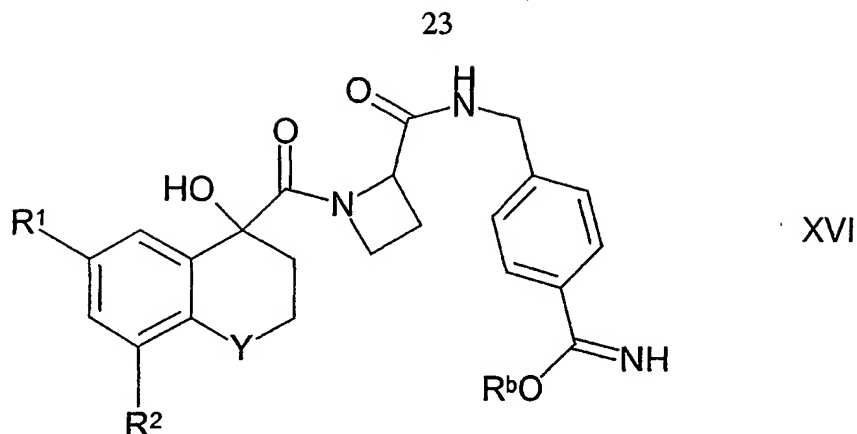


wherein Y, R<sup>1</sup> and R<sup>2</sup> are as hereinbefore defined, with a compound of formula XV,



- 10 wherein R<sup>a</sup> represents H, C<sub>6-10</sub> aryl, C<sub>1-10</sub> alkyl (which latter group is optionally substituted by one or more halo groups) or C<sub>1-3</sub> alkylphenyl, for example at between 40 and 60°C, in the presence of a suitable base (e.g. TEA) and an appropriate organic solvent (e.g. THF, CH<sub>3</sub>CN, DMF or DMSO), optionally by pre-treating the compound of formula XIV with
- 15 gaseous HCl, in the presence of a lower alkyl (e.g. C<sub>1-6</sub> alkyl) alcohol (e.g. ethanol) at, for example, 0°C, to form a compound of formula XVI,





wherein  $R^b$  represents lower (e.g.  $C_{1-6}$ ) alkyl, such as ethyl, and Y,  $R^1$  and  $R^2$  are as hereinbefore defined, which compound may be isolated if desired;

5

(2) for compounds of formula IA in which one of  $D^1$  and  $D^2$  represents  $-OR^7$  (in which  $R^7$  represents H,  $C_{6-10}$  aryl,  $C_{1-10}$  alkyl (which latter group is optionally substituted by one or more halo groups) or  $C_{1-3}$  alkylphenyl), reaction of a corresponding compound of formula IA, in which one of  $D^1$  and  $D^2$  represents  $-C(O)OR^{12}$  (e.g.  $-C(O)O-CH_2CH_2-Si(CH_3)_3$ ) and the other represents H, with a compound of formula XV, as hereinbefore defined, for example at between room and reflux temperature in the presence of an appropriate organic solvent (e.g. THF,  $CH_3CN$ , DMF or DMSO), followed by removal of the  $-C(O)OR^{12}$  group under conditions known to those skilled in the art;

15

(3) for compounds of formula IA in which  $D^1$  or  $D^2$  represents  $R^8$ , reaction of a corresponding compound of formula I or a corresponding compound of formula IA in which  $D^1$  or  $D^2$  (as appropriate) represents H with a compound of formula XVII,

20



XVII

wherein  $L^3$  represents a leaving group, such as halo or *p*-nitrophenoxy, and  $R^8$  is as hereinbefore defined, for example at between 0°C and room temperature in the presence of a suitable base (e.g. NaOH) and an appropriate organic solvent (e.g. THF) and/or water;

5

(4) for compounds of formula IA in which one of  $D^1$  and  $D^2$  represents  $-OR^7$  (wherein  $R^7$  does not represent H), reaction of a corresponding compound of formula IA in which one of  $D^1$  and  $D^2$  represents  $-OH$  with a compound of formula XVIII,

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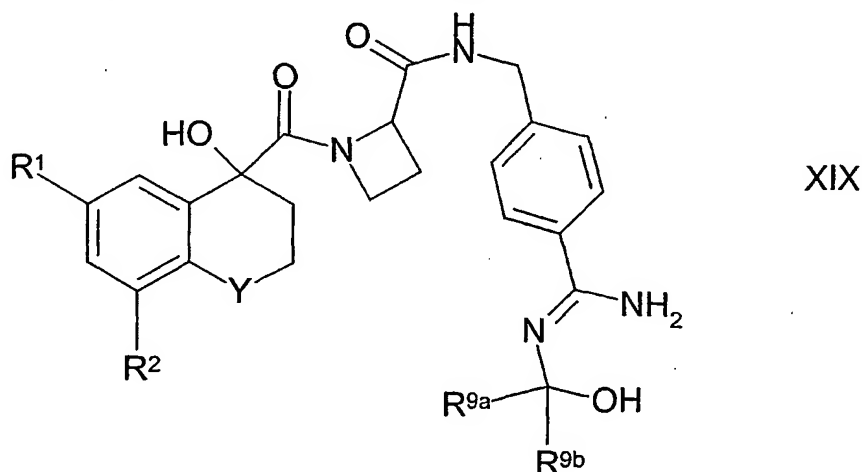
XVIII

wherein  $R^{7a}$  represents  $R^7$  as hereinbefore defined, except that it does not represent H, and  $L^3$  is as hereinbefore defined, for example at between 0°C and reflux temperature, optionally in the presence of an appropriate solvent (e.g. DCM, THF, MeCN or DMF) and a suitable base (e.g.  $Et_3N$  or pyridine); or

15

(5) for compounds of formula IA in which one of  $D^1$  and  $D^2$  represents H and the other represents  $-C(R^{9a})(R^{9b})R^{10}$ , wherein  $R^{10}$  represents  $-OC(O)R^{16a}$ ,  $-OC(O)OR^{17}$  or  $-OC(O)N(R^{18b})R^{17}$ , reaction of a

20



wherein Y, R<sup>1</sup>, R<sup>2</sup>, R<sup>9a</sup> and R<sup>9b</sup> are as hereinbefore defined with a compound of formula XX,



wherein R<sup>c</sup> represents R<sup>16a</sup>, -OR<sup>17</sup> or -N(R<sup>18b</sup>)R<sup>17</sup>, and L<sup>3</sup>, R<sup>16a</sup>, R<sup>17</sup> and R<sup>18b</sup> are as hereinbefore defined, for example under conditions described hereinbefore (e.g. as in step (4) above).

Compounds of formula XIV may be prepared in an analogous fashion to compounds of formula I, for example by the coupling of *p*-cyano-benzylamine with a compound of formula IV, as hereinbefore defined, under conditions such as those described hereinbefore for the synthesis of compounds of formula I (see, for example, process steps (i) and (ii)).

Compounds of formula XIX may be prepared by reaction of a corresponding compound of formula I with an excess of a compound of formula XXI,



wherein R<sup>9a</sup> and R<sup>9b</sup> are as hereinbefore defined, for example under conditions known to those skilled in the art.

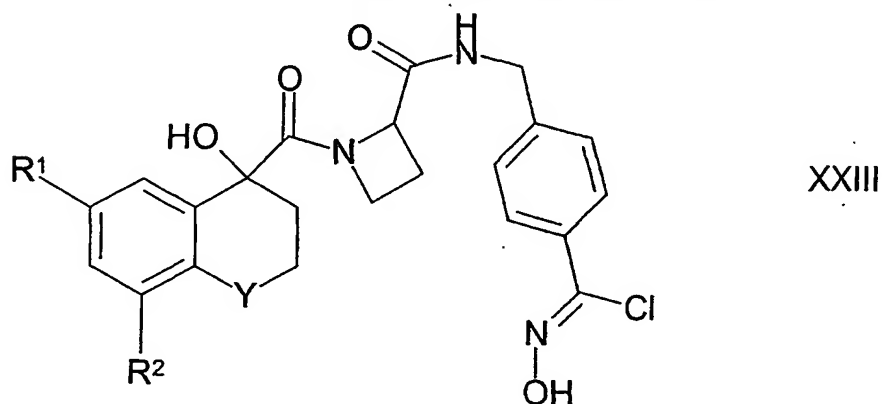
Compounds of formula IA in which D<sup>1</sup>, D<sup>2</sup> and R<sup>3</sup>, together with the amidine group to which they are attached, represent a group IIa may be prepared by reaction of a corresponding compound of formula IA in which D<sup>1</sup> represents OH with a compound of formula XXII,



wherein Hal represents halo (e.g. chloro) and R<sup>4</sup> is as hereinbefore defined, for example at room temperature in the presence of a suitable base (e.g. triethylamine) and an appropriate organic solvent (e.g. dichloromethane), followed by cyclisation of the resultant intermediate,

for example by refluxing in the presence of an appropriate base (e.g. sodium hydride) in a suitable organic solvent (e.g. THF).

Compounds of formula IA in which D<sup>1</sup>, D<sup>2</sup> and R<sup>3</sup>, together with the  
 5 amidine group to which they are attached, represent a group IIb may be prepared by reaction of a corresponding compound of formula XXIII,



wherein Y, R<sup>1</sup> and R<sup>2</sup> are as hereinbefore defined with a compound of  
 10 formula XXIV,



wherein R<sup>b</sup> and R<sup>4</sup> are as hereinbefore defined, for example at room temperature in the presence of a suitable base (e.g. triethylamine) and an appropriate organic solvent (e.g. ethanol).

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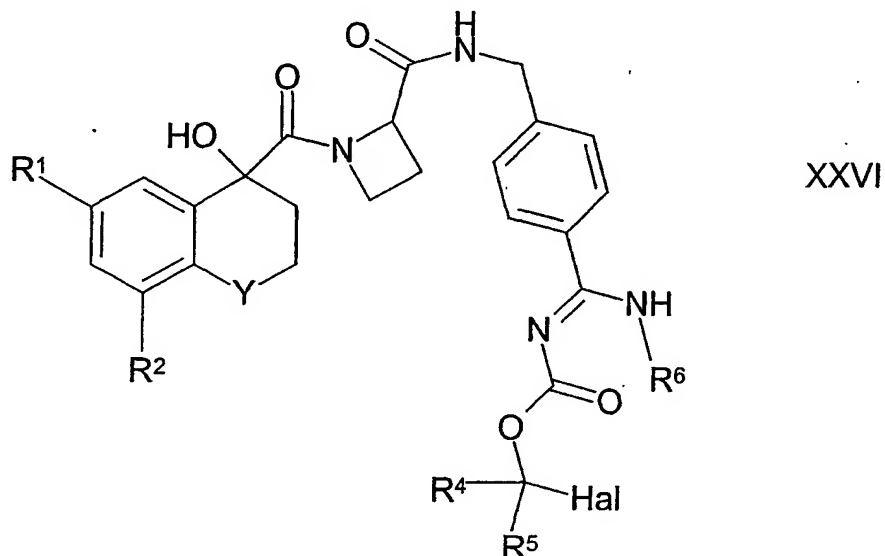
Compounds of formula IA in which D<sup>1</sup>, D<sup>2</sup> and R<sup>3</sup>, together with the amidine group to which they are attached, represent a group IIc may be prepared by reaction of a corresponding compound of formula IA in which D<sup>1</sup> represents OH with a compound of formula XXV,

20



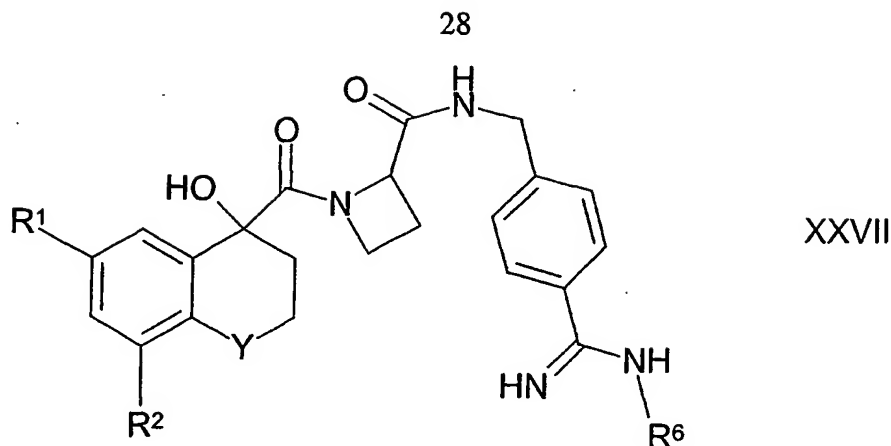
wherein  $R^4$  is as hereinbefore defined, for example at room temperature, followed by oxidation of the resultant intermediate under conditions known to those skilled in the art.

- 5 Compounds of formula IA in which  $D^1$ ,  $D^2$  and  $R^3$ , together with the amidine group to which they are attached, represent a group IId may be prepared by cyclisation of a compound of formula XXVI,



- wherein Hal is as hereinbefore defined (especially iodo) and Y,  $R^1$ ,  $R^2$ ,  $R^4$ ,  $R^5$  and  $R^6$  are as hereinbefore defined, for example at room temperature in the presence of a suitable base (e.g. DIPEA) and an appropriate solvent (e.g. dichloromethane).
- 10

- Compounds of formula XXVI may be prepared by reaction of a compound of formula XXVII,
- 15

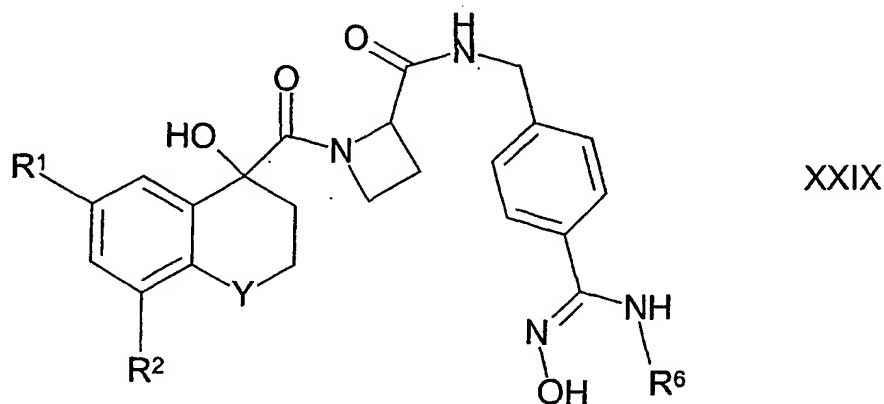


wherein Y, R<sup>1</sup>, R<sup>2</sup> and R<sup>6</sup> are as hereinbefore defined with a compound of formula XXVIII,



- 5 wherein Hal, R<sup>4</sup> and R<sup>5</sup> are as hereinbefore defined for example at room temperature in the presence of a suitable base (e.g. triethylamine) and an appropriate organic solvent (e.g. dichloromethane). It is preferred that, in compounds of formula XXVIII, Hal represents chloro. In such cases, the intermediate of formula XXVI that is formed comprises a chloro group, which, before cyclisation, is preferably converted to an iodo group, using  
10 techniques well known to those skilled in the art (e.g. reaction with NaI).

Compounds of formula IA in which D<sup>1</sup>, D<sup>2</sup> and R<sup>3</sup>, together with the amidine group to which they are attached, represent a group IIe may be  
15 prepared by reaction of a corresponding compound of formula XXIX,



wherein Y, R<sup>1</sup>, R<sup>2</sup> and R<sup>6</sup> are as hereinbefore defined with formaldehyde in water, for example at reflux temperature.

Compounds of formulae XV, XVII, XVIII, XX, XXI, XXII, XXIII, XXIV, XXV, XXVII, XXVIII and XXIX and derivatives thereof, are either commercially available, are known in the literature, or may be obtained either by analogy with the processes described herein, or by conventional synthetic procedures, in accordance with standard techniques, from readily available starting materials using appropriate reagents and reaction conditions (e.g. as described hereinafter).

Compounds of formula IA may be isolated from their reaction mixtures using conventional techniques.

Substituents on the aromatic and/or non-aromatic, carbocyclic and heterocyclic ring(s) in compounds of formulae IA, XIV, XV, XVI, XVII, XVIII, XIX, XX, XXI, XXIII, XXVI, XXVIII and XXIX may be introduced and/or interconverted using techniques well known to those skilled in the art. For example, hydroxy may be alkylated to give alkoxy or acylated to give acyloxy, and alkoxy and acyloxy may be hydrolysed to give hydroxy.

It will be appreciated by those skilled in the art that in the processes described above (in relation to compounds of formulae I and IA) the functional groups of intermediate compounds may need to be protected by protecting groups.

Functional groups that it is desirable to protect include hydroxy, amino, aldehyde, ketone, 2-hydroxycarboxylic acid and carboxylic acid. Suitable

protecting groups for hydroxy include trialkylsilyl or diarylalkylsilyl groups (e.g. *t*-butyldimethylsilyl, *t*-butyldiphenylsilyl or trimethylsilyl) and tetrahydropyranyl. Suitable protecting groups for carboxylic acid include C<sub>1-6</sub> alkyl or benzyl esters. Suitable protecting groups for amino and amidino include *t*-butyloxycarbonyl, benzyloxycarbonyl or 2-trimethylsilylethoxycarbonyl (Teoc). Amidino nitrogens may also be protected by hydroxy or alkoxy groups, and may be either mono- or diprotected. Aldehydes and ketones may be protected as acetals and ketals, respectively, by reacting with e.g. ethylene glycol. 2-Hydroxy carboxylic acids may be protected by condensing with e.g. acetone.

The protection and deprotection of functional groups may take place before or after coupling, or before or after any other reaction in the abovementioned schemes.

Protecting groups may be removed in accordance with techniques that are well known to those skilled in the art and as described hereinafter.

Persons skilled in the art will appreciate that, in order to obtain compounds of formula I, or formula IA, in an alternative, and, on some occasions, more convenient, manner, the individual process steps mentioned hereinbefore may be performed in a different order, and/or the individual reactions may be performed at a different stage in the overall route (i.e. substituents may be added to and/or chemical transformations performed upon, different intermediates to those mentioned hereinbefore in conjunction with a particular reaction). This may negate, or render necessary, the need for protecting groups.



For example, this is particularly true in respect of the synthesis of compounds of formula IA. In this case, D<sup>1</sup> or D<sup>2</sup> groups which do not represent H may be introduced at an earlier stage in the overall synthesis using the process steps described hereinbefore (see, for example, steps (1) to (5) above). Further, in the synthesis of compounds of formula I, the -OH group that is in the  $\alpha$ -position relative to the carboxylic acid in compounds of formulae III and IV may need to be protected prior to the coupling steps described above.

Accordingly, the order and type of chemistry involved will dictate the need, and type, of protecting groups as well as the sequence for accomplishing the synthesis.

The use of protecting groups is fully described in "Protective Groups in Organic Chemistry", edited by J W F McOmie, Plenum Press (1973), and "Protective Groups in Organic Synthesis", 3<sup>rd</sup> edition, T W Greene & P G M Wutz, Wiley-Interscience (1999).

Protected derivatives of compounds of formulae I and IA may be converted chemically to compounds of formulae I and IA using standard deprotection techniques (e.g. hydrogenation).

#### Medical and pharmaceutical use

Compounds of the invention may possess pharmacological activity as such. Compounds of the invention that may possess such activity include, but are not limited to, compounds of formula I.

However, other compounds of the invention (including compounds of formula IA) may not possess such activity, but may be administered parenterally or orally, and thereafter metabolised in the body to form compounds that are pharmacologically active (including, but not limited to, corresponding compounds of formula I). Such compounds (which also includes compounds that may possess some pharmacological activity, but that activity is appreciably lower than that of the "active" compounds to which they are metabolised), may therefore be described as "prodrugs" of the active compounds.

10

Thus, the compounds of the invention are useful because they possess pharmacological activity, and/or are metabolised in the body following oral or parenteral administration to form compounds which possess pharmacological activity. The compounds of the invention are therefore indicated as pharmaceuticals.

15

According to a further aspect of the invention there is thus provided the compounds of the invention for use as pharmaceuticals.

20 In particular, compounds of the invention are potent inhibitors of thrombin either as such and/or (e.g. in the case of prodrugs), are metabolised following administration to form potent inhibitors of thrombin, for example as demonstrated in the tests described below.

25 By "prodrug of a thrombin inhibitor", we include compounds that form a thrombin inhibitor, in an experimentally-detectable amount, and within a predetermined time (e.g. about 1 hour), following oral or parenteral administration.

The compounds of the invention are thus expected to be useful in those conditions where inhibition of thrombin is required.

5 The compounds of the invention are thus indicated in the treatment and/or prophylaxis of thrombosis and hypercoagulability in blood and tissues of animals including man.

It is known that hypercoagulability may lead to thrombo-embolic diseases.

10 Conditions associated with hypercoagulability and thrombo-embolic diseases which may be mentioned include inherited or acquired activated protein C resistance, such as the factor V-mutation (factor V Leiden), and inherited or acquired deficiencies in antithrombin III, protein C, protein S or heparin cofactor II. Other conditions known to be associated with hypercoagulability and thrombo-embolic disease include circulating  
15 antiphospholipid antibodies (Lupus anticoagulant), homocysteinemi, heparin induced thrombocytopenia and defects in fibrinolysis. The compounds of the invention are thus indicated both in the therapeutic and/or prophylactic treatment of these conditions.

20 The compounds of the invention are further indicated in the treatment of conditions where there is an undesirable excess of thrombin without signs of hypercoagulability, for example in neurodegenerative diseases such as Alzheimer's disease.

25 Particular disease states which may be mentioned include the therapeutic and/or prophylactic treatment of venous thrombosis and pulmonary embolism, arterial thrombosis (e.g. in myocardial infarction, unstable angina, thrombosis-based stroke and peripheral arterial thrombosis) and systemic embolism usually from the atrium during arterial fibrillation or

from the left ventricle after transmural myocardial infarction or caused by congestive heart failure.

Moreover, the compounds of the invention are expected to have utility in prophylaxis of re-occlusion (i.e. thrombosis) after thrombolysis, percutaneous trans-luminal angioplasty (PTA) and coronary bypass operations; the prevention of re-thrombosis after microsurgery and vascular surgery in general.

Further indications include the therapeutic and/or prophylactic treatment of disseminated intravascular coagulation caused by bacteria, multiple trauma, intoxication or any other mechanism; anticoagulant treatment when blood is in contact with foreign surfaces in the body such as vascular grafts, vascular stents, vascular catheters, mechanical and biological prosthetic valves or any other medical device; and anticoagulant treatment when blood is in contact with medical devices outside the body such as during cardiovascular surgery using a heart-lung machine or in haemodialysis.

In addition to its effects on the coagulation process, thrombin is known to activate a large number of cells (such as neutrophils, fibroblasts, endothelial cells and smooth muscle cells). Therefore, the compounds of the invention may also be useful for the therapeutic and/or prophylactic treatment of idiopathic and adult respiratory distress syndrome, pulmonary fibrosis following treatment with radiation or chemotherapy, septic shock, septicemia, inflammatory responses, which include, but are not limited to, edema, acute or chronic atherosclerosis such as coronary arterial disease, cerebral arterial disease, peripheral arterial disease, reperfusion damage, and restenosis after percutaneous trans-luminal angioplasty (PTA).

Compounds of the invention that inhibit trypsin and/or thrombin may also be useful in the treatment of pancreatitis.

- 5 According to a further aspect of the present invention, there is provided a method of treatment of a condition where inhibition of thrombin is required which method comprises administration of a therapeutically effective amount of a compound of the invention to a person suffering from, or susceptible to such a condition.

10

The compounds of the invention will normally be administered orally, intravenously, subcutaneously, buccally, rectally, dermally, nasally, tracheally, bronchially, by any other parenteral route or *via* inhalation, in the form of pharmaceutical preparations comprising active compound  
15 either as a free base, or a pharmaceutically acceptable non-toxic organic or inorganic acid addition salt, in a pharmaceutically acceptable dosage form. Depending upon the disorder and patient to be treated and the route of administration, the compositions may be administered at varying doses.

- 20 The compounds of the invention may also be combined and/or co-administered with any antithrombotic agent with a different mechanism of action, such as the antiplatelet agents acetylsalicylic acid, ticlopidine, clopidogrel, thromboxane receptor and/or synthetase inhibitors, fibrinogen receptor antagonists, prostacyclin mimetics and phosphodiesterase  
25 inhibitors, ADP-receptor ( $P_2T$ ) antagonists and inhibitors of carboxypeptidase U (CPU).

The compounds of the invention may further be combined and/or co-administered with thrombolytics such as tissue plasminogen activator

(natural, recombinant or modified), streptokinase, urokinase, prourokinase, anisoylated plasminogen-streptokinase activator complex (APSAC), animal salivary gland plasminogen activators, and the like, in the treatment of thrombotic diseases, in particular myocardial infarction.

5

According to a further aspect of the invention there is thus provided a pharmaceutical formulation including a compound of the invention, in admixture with a pharmaceutically acceptable adjuvant, diluent or carrier.

10 Suitable daily doses of the compounds of the invention in therapeutic treatment of humans are about 0.001-100 mg/kg body weight at peroral administration and 0.001-50 mg/kg body weight at parenteral administration.

15 The compounds of the invention have the advantage that they may, or may be metabolised to compounds that may, be more efficacious, be less toxic, be longer acting, have a broader range of activity, be more potent, produce fewer side effects, be more easily absorbed, or have a better pharmacokinetic profile, than, or have other useful pharmacological,  
20 physical, or chemical, properties over, compounds known in the prior art.

### Biological Tests

The following test procedures may be employed.

#### 25 Test A

##### Determination of Thrombin Clotting Time (TT)

The inhibitor solution (25  $\mu$ L) is incubated with plasma (25  $\mu$ L) for three minutes. Human thrombin (T 6769; Sigma Chem. Co or Hematologic Technologies) in buffer solution, pH 7.4 (25  $\mu$ L, 4.0 NIH units/mL), is

then added and the clotting time measured in an automatic device (KC 10; Amelung).

The thrombin clotting time (TT) is expressed as absolute values (seconds) as well as the ratio of TT without inhibitor ( $TT_0$ ) to TT with inhibitor ( $TT_i$ ). The latter ratios (range 1-0) are plotted against the concentration of inhibitor (log transformed) and fitted to sigmoidal dose-response curves according to the equation

$$y = a/[1 + (x/IC_{50})^s]$$

where:  $a$  = maximum range, i.e. 1;  $s$  = slope of the dose-response curve; and  $IC_{50}$  = the concentration of inhibitor that doubles the clotting time. The calculations are processed on a PC using the software program GraFit Version 3, setting equation equal to: Start at 0, define end = 1 (Erithacus Software, Robin Leatherbarrow, Imperial College of Science, London, UK).

### Test B

#### Determination of Thrombin Inhibition with a Chromogenic, Robotic Assay

The thrombin inhibitor potency is measured with a chromogenic substrate method, in a Plato 3300 robotic microplate processor (Rosys AG, CH-8634 Hombrechtikon, Switzerland), using 96-well, half volume microtitre plates (Costar, Cambridge, MA, USA; Cat No 3690). Stock solutions of test substance in DMSO (72  $\mu$ L), 0.1 - 1 mmol/L, are diluted serially 1:3 (24 + 48  $\mu$ L) with DMSO to obtain ten different concentrations, which are analysed as samples in the assay. 2  $\mu$ L of test sample is diluted with 124  $\mu$ L assay buffer, 12  $\mu$ L of chromogenic substrate solution (S-2366, Chromogenix, Mölndal, Sweden) in assay buffer and finally 12  $\mu$ L of  $\alpha$ -thrombin solution (Human  $\alpha$ -thrombin, Sigma Chemical Co. or Hematologic Technologies) in assay buffer, are added, and the samples

mixed. The final assay concentrations are: test substance 0.00068 - 13.3  $\mu\text{mol/L}$ , S-2366 0.30 mmol/L,  $\alpha$ -thrombin 0.020 NIHU/mL. The linear absorbance increment during 40 minutes incubation at 37°C is used for calculation of percentage inhibition for the test samples, as compared to blanks without inhibitor. The  $\text{IC}_{50}$ -robotic value, corresponding to the inhibitor concentration which causes 50% inhibition of the thrombin activity, is calculated from a log concentration vs. % inhibition curve.

#### 10 Test C

##### Determination of the Inhibition Constant $K_i$ for Human Thrombin

$K_i$ -determinations are made using a chromogenic substrate method, performed at 37°C on a Cobas Bio centrifugal analyser (Roche, Basel, Switzerland). Residual enzyme activity after incubation of human  $\alpha$ -thrombin with various concentrations of test compound is determined at three different substrate concentrations, and is measured as the change in optical absorbance at 405 nm.

Test compound solutions (100  $\mu\text{L}$ ; normally in buffer or saline containing BSA 10 g/L) are mixed with 200  $\mu\text{L}$  of human  $\alpha$ -thrombin (Sigma Chemical Co) in assay buffer (0.05 mol/L Tris-HCl pH 7.4, ionic strength 0.15 adjusted with NaCl) containing BSA (10 g/L), and analysed as samples in the Cobas Bio. A 60  $\mu\text{L}$  sample, together with 20  $\mu\text{L}$  of water, is added to 320  $\mu\text{L}$  of the substrate S-2238 (Chromogenix AB, Mölndal, Sweden) in assay buffer, and the absorbance change ( $\Delta A/\text{min}$ ) is monitored. The final concentrations of S-2238 are 16, 24 and 50  $\mu\text{mol/L}$  and of thrombin 0.125 NIH U/mL.



The steady state reaction rate is used to construct Dixon plots, i.e. diagrams of inhibitor concentration vs.  $1/(\Delta A/\text{min})$ . For reversible, competitive inhibitors, the data points for the different substrate concentrations typically form straight lines which intercept at  $x = -K_i$ .

5

### Test D

#### Determination of Activated Partial Thromboplastin Time (APTT)

APTT is determined in pooled normal human citrated plasma with the reagent PTT Automated 5 manufactured by Stago. The inhibitors are  
10 added to the plasma (10  $\mu\text{L}$  inhibitor solution to 90  $\mu\text{L}$  plasma) and incubated with the APTT reagent for 3 minutes followed by the addition of 100  $\mu\text{L}$  of calcium chloride solution (0.025 M) and APTT is determined by use of the coagulation analyser KC10 (Amelung) according to the instructions of the reagent producer.

15

The clotting time is expressed as absolute values (seconds) as well as the ratio of APTT without inhibitor ( $\text{APTT}_0$ ) to APTT with inhibitor ( $\text{APTT}_i$ ). The latter ratios (range 1-0) are plotted against the concentration of inhibitor (log transformed) and fitted to sigmoidal dose-response curves  
20 according to the equation

$$y = a/[1 + (x/\text{IC}_{50})^s]$$

where:  $a$  = maximum range, i.e. 1;  $s$  = slope of the dose-response curve; and  $\text{IC}_{50}$  = the concentration of inhibitor that doubles the clotting time.

The calculations are processed on a PC using the software program GraFit  
25 Version 3, setting equation equal to: Start at 0, define end = 1 (Erithacus Software, Robin Leatherbarrow, Imperial College of Science, London, UK).

IC<sub>50</sub>APTT is defined as the concentration of inhibitor in human plasma that doubled the Activated Partial Thromboplastin Time.

### Test E

#### 5 Determination of Thrombin Time *ex vivo*

The inhibition of thrombin after oral or parenteral administration of the compounds of formula I, dissolved in ethanol:Solutol™:water (5:5:90), is examined in conscious rats which, one or two days prior to the experiment, are equipped with a catheter for blood sampling from the  
10 carotid artery. On the experimental day blood samples are withdrawn at fixed times after the administration of the compound into plastic tubes containing 1 part sodium citrate solution (0.13 mol per L) and 9 parts of blood. The tubes are centrifuged to obtain platelet poor plasma. The plasma is used for determination of thrombin time or ecarin clotting time  
15 (ECT) as described below.

The citrated rat plasma, 100 µL, is diluted with a saline solution, 0.9%, 100 µL, and plasma coagulation is started by the addition of human thrombin (T 6769, Sigma Chem Co, USA or Hematologic Technologies)  
20 in a buffer solution, pH 7.4, 100 µL, or ecarin (Pentapharm). The clotting time is measured in an automatic device (KC 10, Amelung, Germany).

Where a "prodrug" compound of formula I (e.g. of formula IA) is  
25 administered, concentrations of the appropriate active thrombin inhibitor of formula I in the rat plasma are estimated by the use of standard curves relating the thrombin time or ecarin clotting time in the pooled citrated rat plasma to known concentrations of the corresponding "active" thrombin inhibitor dissolved in saline.

Based on the estimated plasma concentrations of the active thrombin inhibitor (which assumes that thrombin time or ECT prolongation is caused by the aforementioned compound) in the rat, the area under the curve after oral and/or parenteral administration of the corresponding  
5 prodrug compound of formula I is calculated (AUC<sub>pd</sub>) using the trapezoidal rule and extrapolation of data to infinity.

The bioavailability of the active thrombin inhibitor after oral or parenteral  
10 administration of the prodrug is calculated as below:

$$[(\text{AUC}_{\text{pd}}/\text{dose})/(\text{AUC}_{\text{active, parenteral}}/\text{dose})] \times 100$$

where AUC<sub>active, parenteral</sub> represents the AUC obtained after parenteral  
15 administration of the corresponding active thrombin inhibitor to conscious rats as described above.

#### Test F

##### Determination of Thrombin Time in Urine *ex vivo*

20 The amount of the “active” thrombin inhibitor that is excreted in urine after oral or parenteral administration of “prodrug” compounds of the invention, dissolved in ethanol:Solutol™:water (5:5:90), is estimated by determination of the thrombin time in urine *ex vivo* (assuming that thrombin time prolongation is caused by the aforementioned compound).

25

Conscious rats are placed in metabolism cages, allowing separate collection of urine and faeces, for 24 hours following oral administration of compounds of the invention. The thrombin time is determined on the collected urine as described below.

Pooled normal citrated human plasma (100 µL) is incubated with the concentrated rat urine, or saline dilutions thereof, for one minute. Plasma coagulation is then initiated by the administration of human thrombin (T 6769, Sigma Chem Company) in buffer solution (pH 7.4; 100 µL). The clotting time is measured in an automatic device (KC 10; Amelung).

The concentrations of the active thrombin inhibitor in the rat urine are estimated by the use of standard curves relating the thrombin time in the pooled normal citrated human plasma to known concentrations of the aforementioned active thrombin inhibitor dissolved in concentrated rat urine (or saline dilutions thereof). By multiplying the total rat urine production over the 24 hour period with the estimated mean concentration of the aforementioned active inhibitor in the urine, the amount of the active inhibitor excreted in the urine (AMOUNT<sub>pd</sub>) can be calculated.

The bioavailability of the active thrombin inhibitor after oral or parenteral administration of the prodrug is calculated as below:

$$[(\text{AMOUNT}_{\text{pd}}/\text{dose})/(\text{AMOUNT}_{\text{active,parenteral}}/\text{dose})] \times 100$$

where AMOUNT<sub>active,parenteral</sub> represents the amount excreted in the urine after parenteral administration of the corresponding active thrombin inhibitor to conscious rats as described above.

#### Test G

##### Metabolic Activation of Prodrug Compounds *in vitro*

Prodrug compounds of formula IA are incubated at 37°C with liver microsomes or 10 000 g (referring to the centrifuge speed) supernatant

fractions (i.e. s9 fraction) prepared from human or rat liver homogenate. The total protein concentration in the incubations are 1 or 3 mg/mL dissolved in 0.05 mol/L TRIS buffer (pH 7.4), and with the cofactors NADH (2.5 mmol/L) and NADPH (0.8 mmol/L) present. The total  
5 volume of the incubate is 1.2 mL. The initial prodrug concentrations are 5 or 10  $\mu$ mol/L. Samples are collected from the incubate at regular intervals more than 60 minutes after the start of the incubations. Samples (25  $\mu$ L) from the incubate are mixed with an equal volume of human or rat plasma and an appropriate amount of thrombin, and the clotting time  
10 (i.e. thrombin time) is measured on a coagulometer (KC 10; Amelung). The amount of "active" thrombin inhibitor formed is estimated by the use of standard curves relating the thrombin time in pooled citrated human or rat plasma to known concentrations of the corresponding "active thrombin inhibitor".  
15  
The amount of "active" thrombin inhibitor is alternatively, or in addition to the above-mentioned method, estimated by the use of LC-MS.

#### Test H

##### 20 Determination of Plasma Clearance in Rat

Plasma clearance was estimated in male Sprague Dawley rats. The compound was dissolved in water and administered as a subcutaneous bolus injection at a dose of 4  $\mu$ mol/kg. Blood samples were collected at frequent intervals up to 5 hours after drug administration. Blood samples  
25 were centrifuged and plasma was separated from the blood cells and transferred to vials containing citrate (10% final concentration). The ecarin clotting time (ECT) was then determined in each plasma sample by the use of a coagulometer (KC10; Amelung). The plasma concentration in each sample was determined by the use of a standard curve relating the

ECT in pooled citrated plasma samples to known concentrations of the compound. The area under the plasma concentration-time profile was estimated using the log/linear trapezoidal rule and extrapolated to infinite time. Plasma clearance (CL) of the compound was then determined as

5

$$CL = \text{Dose}/AUC$$

The values are reported in mL/min/kg.

#### 10 Test I

##### Determination of *in vitro* Stability of Active Thrombin Inhibitors

Liver microsomes were prepared from Sprague-Dawley rats and human liver samples according to internal SOPs. The compounds were incubated at 37°C at a total microsome protein concentration of 3 mg/mL in a 0.05  
15 mol/L TRIS buffer at pH 7.4, in the presence of the cofactors NADH (2.5 mmol/L) and NADPH (0.8 mmol/L). The initial concentration of compound was 5 or 10 µmol/L. Samples were taken for analysis up to 60 minutes after the start of the incubation. The enzymatic activity in the collected sample was immediately stopped by adding 20% myristic acid at  
20 a volume corresponding to 3.3% of the total sample volume. The concentration of compound remaining (FINAL CONC) in the 60 min. sample was determined by means of LCMS using a sample collected at zero time as reference (START CONC). The % of degraded thrombin inhibitor was calculated as:

25

$$100\% \times \frac{[\text{START CONC}] - [\text{FINAL CONC}]}{[\text{START CONC}]}$$

Test JArterial Thrombosis Model

Vessel damage was induced by applying ferric chloride ( $\text{FeCl}_3$ ) topically to the carotid artery. Rats are anaesthetised with an intraperitoneal  
5 injection of sodium pentobarbital (80 mg/kg; Apoteksbolaget; Umeå, Sweden), followed by continuous infusion (12 mg/kg/h) throughout the experiment. Rat body temperature was maintained at 38°C throughout the experiment by external heating. The experiment started with a 5 minutes control period. Five minutes later, human  $^{125}\text{I}$ -fibrinogen (80 kBq; IM53;  
10 Amersham International, Buckinghamshire, UK) was given intravenously and was used as a marker for the subsequent incorporation of fibrin(ogen) into the thrombus. The proximal end of the carotid artery segment was placed in a plastic tube (6 mm; Silastic®; Dow Corning, MI, USA) opened lengthways, containing  $\text{FeCl}_3$ -soaked (2  $\mu\text{l}$ ; 55% w/w; Merck, Darmstadt,  
15 Germany) filter paper (diameter 3 mm; 1F; Munktell, Grycksbo, Sweden). The left carotid artery was exposed to  $\text{FeCl}_3$  for 10 minutes and was then removed from the plastic tube and soaked in saline. Fifty minutes later, the carotid artery was removed and rinsed in saline. Reference blood samples were also taken for determination of blood  $^{125}\text{I}$  –  
20 activity, 10 minutes after the injection of  $^{125}\text{I}$  –fibrinogen, and at the end of the experiment. The  $^{125}\text{I}$  -activity in the reference blood samples and the vessel segment were measured in a gamma counter (1282 Compugamma; LKB Wallac Oy, Turku, Finland) on the same day as the experiment was performed. The thrombus size was determined as the amount of  $^{125}\text{I}$ -  
25 activity incorporated in the vessel segment in relation to the  $^{125}\text{I}$ -activity in the blood (cpm/mg).

## Examples

The invention is illustrated by way of the following examples. The amino acid Aze is defined as the *S*-isomer if not otherwise specified. The examples were obtained as diastereoisomers if not otherwise specified.

### Example 1

#### (R)-6-Chloro-4-hydroxy-8-methoxy-1,1-dioxothiochromane-4-yl-C(O)-Aze-Pab

10

##### (i) 3-(4-Chloro-2-methoxyphenylthio)propanoic acid, ethyl ester

4-Chloro-2-methoxythiophenol (5.78 g, 33 mmol), triethylamine (5.0 mL, 36 mmol) and ethyl 3-bromopropanoate (4.6 mL, 36 mmol) were dissolved in ethyl acetate (50 mL) and refluxed for 2 h.

15 Water was added and the liquid phases were separated. The aqueous phase was extracted two times with ethyl acetate. The combined organic fractions were washed with 1 M aq. HCl and then concentrated *in vacuo*. Purification by flash chromatography (SiO<sub>2</sub>; heptane : dichloromethane (1:1, 0:1)) afforded the sub-title compound as a yellow solid. Yield: 3.0 g (30%). Estimated purity: 75% (NMR)

20 Bis(4-chloro-2-methoxyphenyl) disulfide (1.7 g, 4.9 mmol) was also isolated (*vide supra*). This material was dissolved in THF : EtOH (80 mL of 1:1) and stirred under N<sub>2</sub>(g). A solution of sodium borohydride (0.38 g, 10 mmol) in 2 M aq. NaOH (2 mL) and water (10 mL) was added. The yellow colour disappeared. After 1 h at room temperature ethyl 3-bromopropanoate (1.3 mL, 10 mmol) was added, resulting in evolution of gas and a white precipitate. After stirring overnight, the solvent was removed *in vacuo* and the residue transferred to a separatory funnel with water and dichloromethane. The organic phase was dried



(MgSO<sub>4</sub>) and concentrated *in vacuo* affording an additional 2.36 g (88%) of the sub-title compound, a yellow oil that solidified after a few hours.

<sup>1</sup>H NMR (400 MHz; CDCl<sub>3</sub>): δ 7.29 (d, 1H), 6.94 (dd, 1H), 6.87 (d, 1H), 4.17 (q, 2H), 3.92 (s, 3H), 3.15 (t, 2H), 2.61 (t, 2H), 1.18 (t, 3H).

5 (ii) 3-(4-Chloro-2-methoxyphenylthio)propanoic acid

To an ice/water-cooled solution of 3-(4-chloro-2-methoxyphenylthio)-propanoic acid, ethyl ester (2.31 g, 8.41 mmol; see step (i) above) in THF (50 mL) was added a solution of lithium hydroxide monohydrate (0.43 g, 10 mmol) in water (25 mL). The temperature was allowed to slowly rise to ambient overnight. THF was removed *in vacuo*, the residue was washed with diethyl ether, acidified with 2 M aq. HCl and then extracted with diethyl ether. The combined ether fractions were extracted with sat. aq. sodium hydrogencarbonate. Careful acidification with aq. HCl  
15 afforded a precipitate that was filtered off and air-dried.

Yield: 1.71 g (82%).

MS(m/z) 245 (M-1).

(iii) 6-Chloro-8-methoxy-4-thiochromanone

20 To a solution/suspension of 3-(4-chloro-2-methoxyphenylthio)propanoic acid (1.46 g, 5.92 mmol; see step (ii) above) in dichloromethane (50 mL) under N<sub>2</sub>(g) was added boron trifluoride dimethyl etherate (1.4 mL, 15 mmol) and trifluoroacetic anhydride (2.1 mL, 15 mmol). After 2.5 h at rt the reaction was quenched with water (2 mL), then  
25 washed with more water. The deep green colour disappeared when it was further washed with sat. aq. sodium hydrogencarbonate. Drying (MgSO<sub>4</sub>) and removal of the solvent *in vacuo* afforded the sub-title compound as a yellow solid.

Yield: 1.19 g (88%)

<sup>1</sup>H NMR (300 MHz; CDCl<sub>3</sub>) δ 7.73 (d, 1H), 6.92 (d, 1H), 3.91 (s, 3H), 3.21 (m, 2H), 2.95 (m, 2H)

(iv) 6-Chloro-4-hydroxy-8-methoxythiochromane-4-yl-carboxylic acid, ethyl ester

To a stirred solution/suspension of 6-chloro-8-methoxy-4-thiochromanone (2.24g; 9.79mmol; see step (iii) above) and zinc iodide (0.078g; 0.24mmol) in dry methylene chloride (20 mL) under N<sub>2</sub>(g) was added trimethylsilyl cyanide (1.35 mL; 10.1 mmol). After 2 days at room temperature, the reaction mixture was slowly added to stirred ice/water-cooled absolute ethanol (50 mL), presaturated with HCl(g) at 0°C. After this addition, the cooling bath was removed and the reaction was monitored using reversed-phase HPLC. After 2 h at rt, the solvents were removed *in vacuo*. The residue was dissolved in THF (40 mL) and H<sub>2</sub>SO<sub>4</sub> (aq.; 0.5 M; 40 mL) and stirred overnight. After concentration under reduced pressure (THF removed), extraction with EtOAc and removal of the solvents *in vacuo*, 3.7 g of crude oil was obtained. Purification using preparative reversed-phase HPLC (acetonitrile:0.1 M aq. ammonium acetate) afforded, after extraction of the appropriate fractions with methylene chloride, the sub-title compound as an oil that slowly crystallized.

Yield: 2.26 g (76%)

<sup>1</sup>H NMR (300 MHz; CDCl<sub>3</sub>) δ 6.80 (d, 1H), 6.73 (d, 1H), 4.20-4.35 (m, 2H), 3.98 (s, 1H), 3.87 (s, 3H), 3.19 (m, 1H), 2.98 (m, 1H), 2.27-2.42 (m, 2H), 1.23 (t, 3H)

(v) 6-Chloro-4-hydroxy-8-methoxythiochromane-4-yl-carboxylic acid

To an ice/water-cooled solution of 6-chloro-4-hydroxy-8-methoxythiochromane-4-yl-carboxylic acid, ethyl ester (2.26 g, 7.46 mmol; see

step (iv) above) in THF (20 mL) was added a solution of lithium hydroxide monohydrate (0.68 g, 16 mmol) in water (10 mL). The temperature was allowed to slowly rise to ambient overnight. THF was removed *in vacuo*, the residue was washed with diethyl ether, acidified with 2 M aq. HCl and then extracted with ethyl acetate. Drying (MgSO<sub>4</sub>) and removal of the solvent *in vacuo* afforded the sub-title compound. Yield: 1.53 g (75%). MS(m/z) 273 (M-1).

- 10 (vi) (R)-6-Chloro-4-hydroxy-8-methoxythiochromane-4-yl-carboxylic acid  
The enantiomers of 6-chloro-4-hydroxy-8-methoxythiochromane-4-yl-carboxylic acid (3.00 g, 10.9 mmol; see step (v) above) were separated using chiral chromatography (Kromasil TBB; heptane:ethyl acetate:formic acid) affording 1.13 g of the faster moving, not wanted *S*-enantiomer, and 1.19 g (79%) of the slower moving sub-title compound. ee=94.6%

<sup>1</sup>H NMR (500 MHz; CD<sub>3</sub>OD) δ 6.97 (d, 1H), 6.83 (d, 1H), 3.83 (s, 3H), 3.07 (m, 1H), 2.98 (m, 1H), 2.40 (m, 1H), 2.27 (m, 1H)

- 20 (vii) (R)-6-Chloro-4-hydroxy-8-methoxy-1,1-dioxothiochromane-4-yl-carboxylic acid

A solution of (R)-6-chloro-4-hydroxy-8-methoxythiochromane-4-yl-carboxylic acid (0.55 g, 2.0 mmol; from step (vi) above) and aqueous hydrogen peroxide (1.0 mL of 35%, 10 mmol) in acetic acid (20 mL) was stirred overnight at ambient temperature. The solvent was removed *in vacuo*, after which the residue was dissolved in water and then freeze dried. Yield: 0.63 g (100%).

MS(m/z) 305 (M-1).

<sup>1</sup>H NMR (400 MHz; CD<sub>3</sub>OD) δ 7.23 (d, 1H), 7.06 (d, 1H), 3.96 (s, 3H), 3.61 (m, 2H), 2.85 (m, 1H), 2.53 (m, 1H).

(viii) (R)-6-Chloro-4-hydroxy-8-methoxy-1,1-dioxothiochromane-4-yl-  
5 C(O)-Aze-Pab(Teoc)

HATU (0.94 g; 2.5 mmol) was added to a solution of (R)-6-chloro-4-hydroxy-8-methoxy-1,1-dioxothiochromane-4-yl-carboxylic acid (0.63 g; 2.05 mmol; see step (vii) above) in DMF (10 mL) at 0°C. After 1 h, a solution of H-Aze-Pab(Teoc) x 2 HCl (1.1 g; 2.5 mmol; see international  
10 patent application WO 98/57932) and *N,N*-diisopropylethylamine (1.4 mL; 8.0 mmol) in DMF (5 mL) was added dropwise. The temperature was allowed slowly to rise to ambient overnight. The solvents were removed *in vacuo* and the residue was purified using reversed-phase HPLC (acetonitrile:0.1 M aq. ammonium acetate). Freeze drying of the  
15 appropriate fractions afforded the sub-title compound as a colourless solid. Yield: 0.50g (37%)  
MS(m/z) 663 (M-1)<sup>-</sup> ; 665 (M+1)<sup>+</sup>

(An alternative method of obtaining (R)-6-chloro-4-hydroxy-8-methoxy-  
20 1,1-dioxothiochromane-4-yl-C(O)-Aze-Pab(Teoc) is as follows: HATU (0.47 g, 1.2 mmol) was added to a solution of (R)-6-chloro-4-hydroxy-8-methoxythiochromane-4-yl-carboxylic acid (0.31 g, 1.1 mmol; see step (vi) above) in DMF (7 mL) at 0°C. After 1 h, a solution of H-Aze-Pab(Teoc) x 2HCl (0.61 g, 1.4 mmol; prepared as described in  
25 international patent application WO 98/57932) and 2,4,6-collidine (0.67 mL, 5.1 mmol) in DMF (9 mL) was added dropwise. After 2 h the reaction mixture was put in a freezer overnight. The solvent was removed *in vacuo*, the residue dissolved in ethyl acetate and then washed with aq. sat. sodium hydrogencarbonate. Purification using reversed-phase HPLC

(acetonitrile : 0.1 M aq. ammonium acetate) afforded, after freeze-drying of the appropriate fractions, the sub-title compound as a colourless solid. To an ice/water-cooled solution of the (*R*)-6-chloro-4-hydroxy-8-methoxythiochromane-4-yl-C(O)-Aze-Pab(Teoc) so obtained (0.25 g, 0.39 mmol), in dichloromethane (20 mL), was added *m*-chloroperbenzoic acid (0.25 g of ~60%, 0.86 mmol), dissolved in a small volume of dichloromethane. The reaction mixture stirred for 2.5 h, then washed with sat. aq. sodium hydrogencarbonate, concentrated *in vacuo* and purified using reversed-phase HPLC (acetonitrile : 0.1 M aq. ammonium acetate). The fractions of interest were concentrated and extracted with ethyl acetate. Drying (MgSO<sub>4</sub>) and removal of the solvent *in vacuo* afforded the sub-title compound. Yield: 0.201 g (76%).)

(ix) (*R*)-6-Chloro-4-hydroxy-8-methoxy-1,1-dioxothiochromane-4-yl-C(O)Aze-Pab x HOAc

To a solution of (*R*)-6-chloro-4-hydroxy-8-methoxy-1,1-dioxothiochromane-4-yl-C(O)-Aze-Pab(Teoc) (0.092 g, 0.14 mmol; see step (viii) above) in THF (15 mL) was added tetrabutylammonium fluoride (0.18 mL of 1.0 M, 0.18 mmol). The solution was stirred at 50°C for 20 h, at 60°C for 5 h, then left at rt overnight. The solvent was removed *in vacuo*. The residue was purified using reversed-phase HPLC (acetonitrile: 0.1 M aq. ammonium acetate) affording the title compound as a white solid, after freeze-drying the appropriate fractions.

Yield: 0.044 g (50%).

MS(*m/z*) 519 (*M*-1)<sup>-</sup>, 521 (*M*+1)<sup>+</sup>.

<sup>1</sup>H NMR (600 MHz; CD<sub>3</sub>OD): (complex due to diastereomers/rotamers) δ 7.76 (d, 0.9H, rotamer); 7.66 (d, 1.1H, rotamer); 7.54 (d, 0.9H, rotamer); 7.47 (d, 1.1H, rotamer); 7.12-7.25 (several peaks, 2H); 5.54 (dd, 0.45H, rotamer); 4.54-4.65 (several peaks, 1.55H); 4.43-4.51

(several peaks, 1H); 4.30 (m, 0.55H, rotamer); 4.09 (m, 0.45H, rotamer); 3.99 (m, 0.45H, rotamer); 3.92 (m, 0.55H, rotamer); 3.90 (s, 3H); 3.76 (m, 0.45H, rotamer); 3.60 (m, 1.1H, rotamer); 3.40 (m, 0.45H, rotamer); 2.70-2.88 (several peaks, 1.45H); 2.46-2.61 (several peaks, 1.55H); 2.28 (m, 0.55H, rotamer); 2.14 (m, 0.45H, rotamer); 1.90 (s, 3H).

<sup>13</sup>C NMR (75 MHz; CD<sub>3</sub>OD): (complex due to diastereomers/rotamers, carbonyl and/or amidine carbons) δ 179.8; 174.2; 173.8; 172.9; 168.1.

## 10 Example 2

### (R)-6,8-Dichloro-4-hydroxy-1,1-dioxothiochromane-4-yl-C(O)-Aze-Pab

#### (i) 3-(2,4-Dichlorophenylthio)propanoic acid, ethyl ester

2,4-Dichlorothiophenol (5.04 g, 28.2 mmol), caesium carbonate (10.1 g, 31 mmol) and ethyl 3-bromopropanoate (4.2 mL, 33 mmol) in acetone (100 mL) were refluxed overnight under N<sub>2</sub>(g). After filtration and concentration under reduced pressure, the residue was partitioned between dichloromethane and water. After separation, the aqueous phase was extracted once with dichloromethane. Drying and removal of the solvent *in vacuo* afforded a liquid residue that contained ~30% (HPLC) of unreacted 2,4-dichlorothiophenol. The mixture was refluxed again with caesium carbonate (4.7 g) and ethyl 3-bromopropanoate (2 mL) in acetonitrile (100 mL) under N<sub>2</sub>(g) for 2 h. After filtration and concentration under reduced pressure, the residue was dissolved in diethyl ether and washed with 2 M aq. sodium hydroxide and water. Drying and removal of the solvent *in vacuo* afforded the sub-title compound.

Yield: 5.85 g (74%).

<sup>1</sup>H NMR (300 MHz; CDCl<sub>3</sub>): δ 7.39 (d, 1H), 7.25 (d, 1H), 7.18 (dd, 1H), 4.13 (q, 2H), 3.15 (t, 2H), 2.61 (t, 2H), 1.13 (t, 3H).

(ii) 3-(2,4-Dichlorophenylthio)propanoic acid

The sub-title compound was prepared using the method described in Example 1(ii) above, starting from 3-(2,4-dichlorophenylthio)propanoic acid, ethyl ester (5.8 g, 20.8 mmol; from step (i) above). Yield: 3.43 g (66%).

MS(m/z) 249 (M-1)<sup>-</sup>.

(iii) 6,8-Dichloro-4-thiochromanone

10 A deep red colour was immediately observed when 3-(2,4-dichlorophenylthio)propanoic acid (1.20 g, 4.78 mmol; from step (ii) above) was added to conc. sulfuric acid (30 mL). After stirring overnight at ambient temperature the reaction mixture was poured into ice. The red colour disappeared. Extraction with diethyl ether, washing with aq., sat. sodium hydrogencarbonate, drying (MgSO<sub>4</sub>) and removal of the solvent *in vacuo* 15 afforded a deep yellow solid.

Yield: 0.93 g (84%).

<sup>1</sup>H NMR (400 MHz; CDCl<sub>3</sub>) δ 8.01 (d, 1H), 7.47 (d, 1H), 3.23 (m, 2H), 2.96 (m, 2H).

20

(iv) 6,8-Dichloro-4-hydroxythiochromane-4-yl-carboxylic acid, ethyl ester

The sub-title compound was prepared according to the method of C.F. Bigge *et al.* in J. Med. Chem., (1993), 36, 1977 using 6,8-dichloro-4-thiochromanone (2.20 g, 9.44 mmol; from step (iii) above). Yield: 25 2.75 g (95%).

<sup>1</sup>H NMR (500 MHz; CDCl<sub>3</sub>) δ 7.30(d, 1H), 7.08 (d, 1H), 4.24-4.35 (m, 2H), 3.23 (m, 1H), 3.04 (ddd, 1H), 2.29-2.41 (m, 2H), 1.25 (t, 3H).

(v) 6,8-Dichloro-4-hydroxythiochromane-4-yl-carboxylic acid

The sub-title compound was prepared using the method described in Example 1(v) above, starting from 6,8-dichloro-4-hydroxychromane-4-yl-carboxylic acid, ethyl ester (2.74 g, 8.92 mmol; from step (iv) above).

5 Yield: 2.17 g (87%).

MS(m/z) 277 (M-1)<sup>-</sup>.

<sup>1</sup>H NMR (400 MHz; CDCl<sub>3</sub>) δ 7.34 (d, 1H), 7.21 (d, 1H), 3.21 (m, 1H), 3.12 (ddd, 1H), 2.35-2.50 (m, 2H), 2.13 (s, 1H).

10 (vi) (R)-6,8-Dichloro-4-hydroxythiochromane-4-yl-carboxylic acid

The enantiomers of 6,8-dichloro-4-hydroxychromane-4-yl-carboxylic acid (2.17 g, 7.77 mmol; from step (v) above) were separated using chiral chromatography (Kromasil TBB; heptane : ethyl acetate : formic acid) affording 0.74 g of the faster moving, not wanted *S*-enantiomer, and

15 0.72 g (66%) of the slower moving sub-title compound. ee=98.4%

(vii) (R)-6,8-Dichloro-4-hydroxythiochromane-4-yl-C(O)-Aze-Pab(Teoc)

The sub-title compound was prepared analogously to the method described in Example 1(viii) above, starting from (*R*)-6,8-dichloro-4-hydroxy-  
20 chromane-4-yl-carboxylic acid (0.19 g, 0.68 mmol; from step (vi) above) and H-Aze-Pab(Teoc) (0.37 g, 0.82 mmol; prepared as described in international patent application WO 98/57932). Yield: 0.28 g (65%).

MS(m/z) 635 (M-1)<sup>-</sup>, 637 (M+1)<sup>+</sup>.

25 (viii) (R)-6, 8-Dichloro-4-hydroxy-1,1-dioxothiochromane-4-yl-C(O)-Aze-Pab(Teoc)

The sub-title compound was prepared analogously to the method described in Example 1(viii) (alternative preparation, second step) above, starting from (*R*)-6,8-dichloro-4-hydroxy-thiochromane-4-yl-C(O)-Aze-Pab(Teoc)



(0.100 g, 0.157 mmol; from step (vii) above) and *m*-chloroperbenzoic acid (0.125 g of 60%; 0.72 mmol). Yield: 0.040 g (38%).

MS(*m/z*) 667 (*M*-1)<sup>-</sup>, 669 (*M*+1)<sup>+</sup>.

5 (ix) (*R*)-6,8-Dichloro-4-hydroxy-1,1-dioxothiochromane-4-yl-C(O)-Aze-Pab x CF<sub>3</sub>COOH

Trifluoroacetic acid (1.0 mL) was added to a stirred ice/water-cooled solution of (*R*)-6,8-dichloro-4-hydroxy-1,1-dioxothiochromane-4-yl-C(O)-Aze-Pab(Teoc) (40 mg, 60 μmol; from step (viii) above) in  
10 dichloromethane (10 mL). The cooling bath was removed after 30 min. After 1 h at rt, acetonitrile (30 mL) was added and the solvents were carefully removed under reduced pressure. The residue was dissolved in water and then freeze-dried.

Yield: 27 mg (68%).

15 MS(*m/z*) 525 (*M*-1)<sup>-</sup>, 523 (*M*+1)<sup>+</sup>.

<sup>1</sup>H NMR (400 MHz; CD<sub>3</sub>OD): (complex due to diastereomers/rotamers) δ 8.74 (m, 1H), 7.45-7.80 (several peaks, 6H), 5.53 (dd, 0.4H, rotamer), 4.35-4.88 (several peaks, 3.8H), 4.09 (m, 0.4H, rotamer), 4.00 (m, 0.4H, rotamer), 3.87 (m, 0.4H, rotamer), 3.68 (m, 1.2H, rotamer), 3.45  
20 (ddd, 0.4H); 2.78-2.90 (m, 1H), 2.73 (m, 0.4H, rotamer); 2.46-2.64 (several peaks, 1.6H); 2.31 (m, 0.6H, rotamer); 2.15 (m, 0.4H, rotamer).

<sup>13</sup>C NMR (100 MHz; CD<sub>3</sub>OD): δ 174.2; 173.4; 168.2.

Example 3

25 (*R*)-6-Chloro-4-hydroxy-1,1-dioxothiochromane-4-yl-C(O)-Aze-Pab

(i) 6-Chloro-4-hydroxythiochromane-4-yl-carboxylic acid, methyl ester

The sub-title compound was prepared according to the method of C.F. Bigge *et al.* in J. Med. Chem., (1993), 36, 1977 from 6-chloro-4-

thiochromanone (1.74 g, 8.76 mmol) using methanol instead of ethanol.

Yield: 0.625 g (28%).

<sup>1</sup>H NMR (500 MHz; CDCl<sub>3</sub>) δ 7.07-7.15 (several peaks, 3H), 3.93 (s, 1H), 3.93 (s, 3H), 3.23 (ddd, 1H), 2.99 (ddd, 1H), 2.42 (ddd, 1H), 2.34 (ddd, 1H).

(ii) 6-Chloro-4-hydroxythiochromane-4-yl-carboxamide

The sub-title compound was obtained as the main product in the previous step and isolated using reversed phase HPLC (acetonitrile : 0.1 M aq. ammonium acetate).

Yield: 1.49 g (70%).

<sup>1</sup>H NMR (500 MHz; CDCl<sub>3</sub>) δ 7.35 (d, 1H), 7.15 (dd, 1H), 7.09 (d, 1H), 6.32 (bs, 1H), 5.90 (bs, 1H), 3.86 (bs, 1H), 3.19 (ddd, 1H), 3.03 (ddd, 1H), 2.40 (ddd, 1H), 2.30 (ddd, 1H).

(iii) 6-Chloro-4-hydroxythiochromane-4-yl-carboxylic acid

Potassium hydroxide (5.1 g, 91 mmol) and 6-chloro-4-hydroxythiochromane-4-yl-carboxamide (1.36 g, 5.58 mmol; from step (ii) above) were dissolved in water (25 mL) and isopropanol and refluxed for 2 d.

The isopropanol was removed under reduced pressure and the aqueous residue was washed with diethyl ether. Acidification with aq. HCl, extraction with ethyl acetate, and then drying and concentration *in vacuo* afforded the sub-title compound. Yield: 0.87 g (64%).

MS(m/z) 243 (M-1).

Alternative method: The sub-title compound was also obtained using the method in Example 1(v) above, starting from 6-chloro-4-hydroxythiochromane-4-yl-carboxylic acid, methyl ester (0.63 g, 2.4 mmol; from step (i) above). Yield: 0.55 g (92%).

(iv) (R)-6-Chloro-4-hydroxythiochromane-4-yl-carboxylic acid

The enantiomers of 6-chloro-4-hydroxythiochromane-4-yl-carboxylic acid (1.4 g, 5.7 mmol; from step (iii) above) were separated using chiral chromatography (Kromasil TBB; heptane : ethyl acetate : formic acid) yielding 0.500 g (71 %) of the slower moving sub-title compound.

(v) (R)-6-Chloro-4-hydroxythiochromane-4-yl-C(O)-Aze-Pab(Teoc)

The sub-title compound was prepared analogously to the method described in Example 1(viii) (alternative preparation, first step) above, starting from (R)-6-chloro-4-hydroxythiochromane-4-yl-carboxylic acid (0.178 g, 0.727 mmol; from step (iv) above) and H-Aze-Pab(Teoc) (0.36 g, 0.80 mmol; prepared as described in international patent application WO 98/57932). Yield: 0.31 g (71 %). MS(m/z) 601 (M-1)<sup>-</sup>; 603 (M+1)<sup>+</sup>.

(vi) (R)-6-Chloro-4-hydroxy-1,1-dioxothiochromane-4-yl-C(O)-Aze-Pab(Teoc)

The sub-title compound was prepared analogously to the method described in Example 1(viii) (alternative preparation, second step) above, starting from (R)-6-chloro-4-hydroxythiochromane-4-yl-C(O)-Aze-Pab(Teoc) (0.113 g, 0.187 mmol; from step (v) above) and *m*-chloroperbenzoic acid (0.16 g of ~60 %, 0.56 mmol). Yield: 0.029 g (27 %). MS(m/z) 633 (M-1)<sup>-</sup>; 635 (M+1)<sup>+</sup>.

(vii) (R)-6-Chloro-4-hydroxy-1,1-dioxothiochromane-4-yl-C(O)-Aze-Pab x HOAc

The title compound was prepared analogously to the method described in Example 1(ix) above, starting from (R)-6-chloro-4-hydroxy-1,1-dioxothiochromane-4-yl-C(O)-Aze-Pab(Teoc) (0.029 g, 0.046 mmol; from step

(vi) above) and tetrabutylammonium fluoride (0.080 mL of 1.0 M, 0.080 mmol).

Yield: 0.023 g (91%).

MS(m/z) 489 (M-1)<sup>-</sup>, 491 (M+1)<sup>+</sup>.

5 <sup>1</sup>H NMR (400 MHz; CD<sub>3</sub>OD): (complex due to diastereomers/rotamers) δ 7.47-7.84 (several peaks, 7H); 5.55 (dd, 0.3H, rotamer); 4.43-4.66 (several peaks, 2.7H); 4.30 (m, 0.7H, rotamer); 4.09 (m, 0.5H, rotamer); 4.00 (m, 0.5H, rotamer); 3.77 (ddd, 0.5H, rotamer); 3.54-3.67 (several peaks, 1.3H); 3.43 (ddd, 0.5H, rotamer); 2.52-2.99 (several peaks, 3H);  
10 2.24-2.33 (m, 1H); 2.10-2.20 (m, 0.7H), rotamer), 1.92 (s, 3H); 1.75-1.85 (m, 0.3H), rotamer).

<sup>13</sup>C NMR (100 MHz; CD<sub>3</sub>OD): (complex due to diastereomers/rotamers, carbonyl and/or amidine carbons) δ 173.5; 172.9; 168.1.

15 Example 4

(R)-6-Chloro-4-hydroxy-8-methoxy-1,1-dioxothiochromane-4-yl-C(O)-Aze-Pab(OBn)

(i) (R)-6-Chloro-4-hydroxy-8-methoxy-1,1-dioxothiochromane-4-yl-C(O)-Aze-Pab(OBn)(Teoc)

20 (R)-6-Chloro-4-hydroxy-8-methoxy-1,1-dioxothiochromane-4-yl-C(O)-Aze-Pab-(Teoc) (0.040 g, 0.060 mmol; see Example 1(viii) above) and O-benzylhydroxylamine x HCl (0.047 g, 0.29 mmol) in THF (15 mL) were heated to 60°C for 2d. After concentration under reduced pressure,  
25 the residue was dissolved in ethyl acetate and washed with water and brine. Drying (MgSO<sub>4</sub>) and removal of the solvent *in vacuo* afforded a colourless solid residue. Yield: 0.046 g (99%).

MS(m/z) 771 (M+1)<sup>+</sup>, 793 (M+23)<sup>+</sup>.

(ii) (R)-6-Chloro-4-hydroxy-8-methoxy-1,1-dioxothiochromane-4-yl-C(O)-Aze-Pab(OBn)

A solution of (R)-6-chloro-4-hydroxy-8-methoxy-1,1-dioxothiochromane-4-yl-C(O)-Aze-Pab(OBn)(Teoc) (0.046 g, 0.060 mmol; from step (i) above) and tetrabutylammonium fluoride (0.20 mL of 1.0 M, 0.20 mmol) in acetonitrile (3 mL) was heated to 60°C for 3 h. After concentration under reduced pressure, the residue was dissolved in ethyl acetate and washed with water and brine. Evaporation of the solvent *in vacuo* afforded only 0.019 g. The aqueous fractions were freeze dried, combined with the evaporation residue from the organic phase and purified using reversed-phase HPLC (acetonitrile : 0.1 M aq. ammonium acetate) to afford, after freeze drying the appropriate fractions, the title compound as a colourless solid.

Yield: 0.009 g (24%).

MS(m/z) 625 (M-1)<sup>-</sup>, 627 (M+1)<sup>+</sup>.

<sup>1</sup>H NMR (400 MHz; CDCl<sub>3</sub>): δ 7.59 (d, 2H), 7.23-7.45 (several peaks, 7H), 6.99 (d, 1H), 6.81 (d, 1H), 5.12 (s, 2H); 4.90 (dd, 1H), 4.85 (bs, 2H), 4.40-4.54 (AB-part of ABX-system, 2H), 4.09 (m, 1H), 3.97 (s, 3H), 3.84 (ddd, 1H), 3.28 (ddd, 1H), 3.09 (m, 1H), 2.79 (ddd, 1H), 2.63 (m, 1H), 2.32 (m, 1H), 2.21 (m, 1H).

<sup>13</sup>C NMR (100 MHz; CDCl<sub>3</sub>): (carbonyl and/or amidine carbons) δ 169.1, 158.3, 151.6.

Example 5

25 (R)-6-Chloro-4-hydroxy-8-methoxy-1,1-dioxothiochromane-4-yl-C(O)-Aze-Pab(C(O)-O-cyclopentyl)

Sodium hydroxide (aq.; 0.5 mL of 2.0 M, 1.0 mmol) was added to a solution of (R)-6-chloro-4-hydroxy-8-methoxy-1,1-dioxothiochromane-4-yl-C(O)-Aze-Pab x HOAc (21 mg, 40 μmol; see Example 1 above) and

cyclopentyl chloroformate (11 mg, 74  $\mu$ mol) in dichloromethane, and the mixture was stirred at rt for 3 h. After dilution with water, the resultant mixture was extracted with dichloromethane. The combined organic fractions were concentrated *in vacuo* and purified using flash chromatography (SiO<sub>2</sub>; dichloromethane, then with EtOAc/MeOH 95/5).  
The fractions of interest were concentrated under reduced pressure, then dissolved in water/acetonitrile and freeze dried to afford the title compound. Yield: 16 mg (60%).

MS(m/z) 631 (M-1)<sup>-</sup>, 633 (M+1)<sup>+</sup>.

<sup>1</sup>H NMR (300 MHz; CDCl<sub>3</sub>): (minor rotamer (~10%) not reported)  $\delta$  7.75 (d, 2H), 7.49 (t, 1H), 7.23 (d, 2H), 6.92 (d, 1H), 6.85 (d, 1H), 5.14 (m, 1H); 4.82 (broad dd, 2H), 4.34-4.58 (AB-part of ABX-system, 2H), 4.15 (m, 1H), 3.97 (bs, 1H), 3.92 (s, 3H), 3.71 (ddd, 1H), 3.39 (ddd, 1H), 3.24 (m, 1H), 2.79 (ddd, 1H), 2.56 (m, 1H), 2.22-2.38 (several peaks, 2H), 1.51-2.00 (several peaks, 8H).  
<sup>13</sup>C NMR (100 MHz; CDCl<sub>3</sub>): (carbonyl and/or amidine carbons)  $\delta$  173.1, 171.1, 169.3, 158.2.

#### Example 6

(R)-6,8-Dichloro-4-hydroxy-1,1-dioxothiochromane-4-yl-C(O)-Aze-Pab(OMe)

(i) (R)-6,8-Dichloro-4-hydroxy-1,1-dioxothiochromane-4-yl-C(O)-Aze-Pab(OMe)(Teoc)

(R)-6,8-Dichloro-4-hydroxy-1,1-dioxothiochromane-4-yl-C(O)-Aze-Pab(Teoc) (0.046 g, 0.069 mmol; from Example 2(viii) above) and *O*-methyl-hydroxylamine x HCl (0.043 g, 0.51 mmol) in THF (5 mL) were refluxed overnight. After concentration under reduced pressure, the residue was dissolved in ethyl acetate and washed with water and brine.

Drying ( $\text{MgSO}_4$ ) and removal of the solvent *in vacuo* afforded a colourless solid residue. Yield: 0.042 g (87%).

(ii) (R)-6,8-Dichloro-4-hydroxy-1,1-dioxothiochromane-4-yl-C(O)-Aze-

5 Pab(OMe)

Trifluoroacetic acid (1.0 mL) was added to a stirred ice/water-cooled solution of (R)-6,8-dichloro-4-hydroxy-1,1-dioxothiochromane-4-yl-C(O)-Aze-Pab(OMe)(Teoc) (0.042 g, 0.060 mmol; from step (i) above) in dichloromethane (3 mL). The cooling bath was removed after 1 h. After  
10 3 h at rt, acetonitrile was added and the solvents were carefully removed under reduced pressure. The crude product was purified using reversed-phase HPLC (acetonitrile : 0.1 M aq. ammonium acetate) to afford, after freeze drying the appropriate fractions, the title compound as a colourless solid.

15 Yield: 0.018 g (54%).

MS(m/z) 553 ( $\text{M}-1$ )<sup>-</sup>, 555 ( $\text{M}+1$ )<sup>+</sup>.

<sup>1</sup>H NMR (400 MHz;  $\text{CDCl}_3$ ): (minor rotamer (~10%) not reported)  $\delta$  7.57 (d, 2H), 7.51 (d, 1H), 7.10-7.55 (several peaks, 4H), 4.96 (bs, 2H); 4.94 (dd, 1H), 4.38-4.54 (AB-part of ABX-system, 2H), 4.18 (m, 1H), 3.91  
20 (s, 3H), 3.76 (ddd, 1H), 3.43 (ddd, 1H), 3.23 (m, 1H), 2.84 (ddd, 1H), 2.58 (m, 1H), 2.27-2.43 (several peaks, 2H).

<sup>13</sup>C NMR (100 MHz;  $\text{CDCl}_3$ ): (carbonyl and/or amidine carbons)  $\delta$  172.7, 169.1, 151.5.

Example 76-Chloro-8-difluoromethoxy-4-hydroxy-1,1-dioxothiochromane-4-yl-C(O)-  
Aze-Pab5 (i) 6-Chloro-8-hydroxy-4-thiochromanone, ethylene ketal

To a solution of 6-chloro-8-methoxy-4-thiochromanone (7.64 g, 33.4 mmol; see Example 1(iii) above) in benzene (300 mL) under nitrogen was added ethylene glycol (9.34 mL, 167 mmol) and PTSA•H<sub>2</sub>O (catalytic amount). The mixture was heated to reflux with a Dean-Stark trap for 48

10 h. In order to drive the reaction to completion it was necessary to add molecular sieves (type 4 Å, 8-12 mesh) to the trap. The reaction was cooled to room temperature, solid NaHCO<sub>3</sub> (0.10 g) was added and the mixture was diluted with EtOAc (200 mL). The organic layer was washed with saturated NaHCO<sub>3</sub> (2 x 150 mL), brine (150 mL), dried

15 (Na<sub>2</sub>SO<sub>4</sub>), filtered, and concentrated *in vacuo* to afford 9.03 g (ca. 99%) of the intermediate ketal as a crude yellow solid. Ethanethiol (24.7 mL, 334 mmol) was added to a mixture of the above ketal (13.0 g, 47.7 mmol) in DMF (250 mL) under nitrogen and cooled to 0°C. Sodium hydride (8.0 g, 334 mmol) was added slowly and the reaction stirred at 0°C for 15

20 minutes, then placed in an oil bath and heated to 135°C for 1.5 h. Once the reaction had cooled to room temperature, it was poured into an ice-water mixture (300 g) containing saturated NH<sub>4</sub>Cl solution (70 mL). The aqueous layer was extracted with EtOAc (2 x 150 mL) and the combined organic extracts were washed with brine (200 mL), dried (Na<sub>2</sub>SO<sub>4</sub>),

25 filtered, and concentrated *in vacuo*. The brown residue was chromatographed on silica gel eluting with CH<sub>2</sub>Cl<sub>2</sub> to afford 11.3 g (92%) of the sub-title compound as a yellow solid.

<sup>1</sup>H NMR (300 MHz, CD<sub>3</sub>OD) δ 7.05 (s, 1H), 6.68 (s, 1H), 4.05-4.20 (m, 4H), 3.07 (t, *J* = 8.0 Hz, 2H), 2.13 (t, *J* = 8.0 Hz, 2H).



(ii) 6-Chloro-8-difluoromethoxy-4-thiochromanone, ethylene ketal

A solution of 6-chloro-8-hydroxy-4-thiochromanone, ethylene ketal (6.0 g, 23.2 mmol; from step (i) above) in 2-propanol (100 mL) and 30% KOH (50 mL) was heated to reflux and Freon 22 was bubbled directly into the reaction for 50 min. at <2.0 psi. The mixture was cooled to room temperature, partitioned between CH<sub>2</sub>Cl<sub>2</sub> and H<sub>2</sub>O (2:1, 450 mL) and the aqueous layer was extracted with CH<sub>2</sub>Cl<sub>2</sub> (150 mL). The combined organic layers were washed with brine (150 mL), dried (Na<sub>2</sub>SO<sub>4</sub>), filtered, and concentrated *in vacuo*. The resulting residue was chromatographed on silica gel eluting with CH<sub>2</sub>Cl<sub>2</sub> to afford 7.2 g (78%) of the sub-title compound as a yellow solid.

<sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ 7.45 (s, 1H), 7.10 (s, 1H), 6.45 (t, *J* = 74 Hz, 1H), 4.09-4.25 (m, 4H), 3.17 (t, *J* = 7.5 Hz, 2H), 2.17 (t, *J* = 7.5 Hz, 2H).

(iii) 6-Chloro-8-difluoromethoxy-4-thiochromanone

A solution of 6-chloro-8-difluoromethoxy-4-thiochromanone, ethylene ketal (5.11 g, 16.6 mmol; from step (ii) above) in THF (105 mL) and 0.5 N HCl (70 mL) was stirred at room temperature for 96 h. The reaction was neutralized with saturated NaHCO<sub>3</sub> (150 mL) and extracted with CH<sub>2</sub>Cl<sub>2</sub> (2 x 100 mL). The combined organic extracts were washed with brine (100 mL), dried (Na<sub>2</sub>SO<sub>4</sub>), filtered, and concentrated *in vacuo* to afford 4.40 g (*ca.* 100%) of the sub-title compound as a yellow solid which was used without further purification.

<sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ 8.00 (s, 1H), 7.30 (s, 1H), 6.56 (t, *J* = 74 Hz, 1H), 3.24 (t, *J* = 7.5 Hz, 2H), 3.00 (t, *J* = 7.5 Hz, 2H).

(iv) 6-Chloro-8-difluoromethoxy-4-methylenethiochromane

A solution of 6-chloro-8-difluoromethoxy-4-thiochromanone (2.5 g, 9.40 mmol; from step (iii) above) in anhydrous Et<sub>2</sub>O (150 mL) was cooled to 0°C under nitrogen. Tebbe Reagent (21.0 mL of 0.5M in toluene, 10.3 mmol) was added dropwise *via* syringe, and the mixture was stirred at 0°C for 2 h and then stirred at room temperature for 1 h. The reaction was cooled to 0°C and slowly quenched with MeOH (*ca.* 50 mL) until the solution solidified. The mixture was diluted with EtOAc/H<sub>2</sub>O (4:1, 100 mL) and filtered through a pad of Celite. The Celite was washed with EtOAc (800 mL) and the filtrate was concentrated *in vacuo*. The crude dark red oil was chromatographed on silica gel eluting with Hex:EtOAc (4:1) to afford 0.88 g (36%) of sub-title compound as a dark yellow oil.

<sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) δ 7.40 (s, 1H), 7.05 (s, 1H), 6.46 (t, *J* = 75 Hz, 1H), 5.53 (s, 1H), 5.10 (s, 1H), 3.06 (t, *J* = 7.5 Hz, 2H), 2.80 (t, *J* = 7.5 Hz, 2H).

(v) (R)-6-Chloro-8-difluoromethoxy-4-hydroxy-4-hydroxymethylthiochromane

2-Methyl-2-propanol (120 mL), H<sub>2</sub>O (120 mL), and AD-mix-β (13.3 g) were combined together and cooled to 0°C. Concurrently, 6-chloro-8-difluoromethoxy-4-methylenethiochromane (2.65 g, 10.1 mmol; from step (iv) above) was cooled to -5°C in an ice/MeOH bath. Once sufficiently cooled, the AD-mix-β solution was added and stirred at <0°C for 2 h, then at room temperature for 4 h. The reaction was quenched with solid sodium sulfite (*ca.* 5.0 g) resulting in a clear solution which was then partitioned between EtOAc/H<sub>2</sub>O (1:1, 300 mL) and further extracted with EtOAc (100 mL). The organic extracts were combined and washed with 5 M H<sub>2</sub>SO<sub>4</sub> (75 mL), saturated NaHCO<sub>3</sub> (150 mL), and brine (150 mL), dried (Na<sub>2</sub>SO<sub>4</sub>), filtered, and concentrated *in vacuo* to afford 2.99 g (*ca.*

100%) of the sub-title compound as a yellow solid which was used without further purification.

<sup>1</sup>H NMR (300 MHz, CD<sub>3</sub>OD) δ 7.55 (s, 1H), 7.08 (s, 1H), 6.78 (t, *J* = 75 Hz, 1H), 3.70 (d, *J* = 12 Hz, 1H), 3.50 (d, *J* = 12 Hz, 1H), 2.97-  
5 3.20 (m, 2H), 2.46-2.55 (m, 1H), 1.91-2.02 (m, 1H).

(vi) (R)-6-Chloro-8-difluoromethoxy-4-hydroxythiochromane-4-yl-carboxylic acid

Sodium carbonate (0.72 g, 6.7 mmol) and 5% Pt/C (1.50 g, 75% weight  
10 of reactant) were added to a solution of (R)-6-chloro-8-difluoromethoxy-4-hydroxy-4-hydroxymethylthiochromane (2.0 g, 6.7 mmol; from step (v) above) in H<sub>2</sub>O (180 mL). The mixture was vigorously stirred at *ca.* 95°C for 20-40 h with air being directly bubbled into the reaction mixture. The reaction was cooled to room temperature and filtered through a pad of  
15 Celite with saturated Na<sub>2</sub>CO<sub>3</sub> solution (*ca.* 100 mL). The resulting liquid was slowly acidified with 2 N HCl (*ca.* 200 mL) and extracted with EtOAc (2 x 100 mL). The combined organic layers were washed with brine (75 mL), dried (Na<sub>2</sub>SO<sub>4</sub>), filtered, and concentrated *in vacuo* to afford 1.56 g (*ca.* 75%) of the sub-title compound as a brown oil which  
20 was used without further purification.

<sup>1</sup>H NMR (300 MHz, CD<sub>3</sub>OD) δ 7.24 (s, 1H), 7.12 (s, 1H), 6.81 (t, *J* = 75 Hz, 1H), 3.04-3.19 (m, 2H), 2.40-2.50 (m, 1H), 2.25-2.35 (m, 1H).

(vii) (R)-6-Chloro-8-difluoromethoxy-4-hydroxy-1,1-dioxothiochromane-4-yl-carboxylic acid

25 A mixture of (R)-6-chloro-8-difluoromethoxy-4-hydroxythiochromane-4-yl-carboxylic acid (0.50 g, 1.6 mmol; from step (vi) above), wet alumina (prepared by adding H<sub>2</sub>O (1 mL) to alumina (5.0 g) and shaking until a free flowing powder was obtained, *ca.* 0.82 g, 8.0 mmol), and Oxone®

(4.9 g, 8.0 mmol) in  $\text{CHCl}_3$  (19 mL) was heated to 40-45°C for 20 h. The reaction was cooled to room temperature, filtered through a Büchner funnel with EtOAc (50 mL), and concentrated *in vacuo* to afford 0.54 g (98%) of the sub-title compound as a brown foam which was used without  
5 further purification.

$^1\text{H}$  NMR (300 MHz,  $\text{CD}_3\text{OD}$ )  $\delta$  7.44 (s, 2H), 6.94 (t,  $J = 75$  Hz, 1H), 3.61-3.75 (m, 2H), 2.81-2.91 (m, 1H), 2.52-2.62 (m, 1H).

(viii) (R)-6-Chloro-8-difluoromethoxy-4-hydroxy-1,1-dioxothiochromane-4-yl-C(O)-Aze-Pab(Teoc)  
10

To a solution of (R)-6-chloro-8-difluoromethoxy-4-hydroxy-1,1-dioxothiochromane-4-yl-carboxylic acid (0.54 g, 1.6 mmol; from step (vii) above) in DMF (20 mL) under argon at 0°C was added H-Aze-Pab(Teoc) x HCl (0.92 g, 2.1 mmol), PyBOP (0.91 g, 1.7 mmol), and Hünig's Base  
15 (0.7 mL, 4.0 mmol). The reaction was stirred at 0°C for 2 h and then at room temperature for 20 h. The mixture was concentrated under reduced pressure and the residue was chromatographed twice on silica gel, eluting first with  $\text{CHCl}_3$ :EtOH (10:1) and then with EtOAc:EtOH (10:1) to afford 0.44 g (40%) of the sub-title compound as a yellow crystalline solid.

20 mp 135-145°C

$^1\text{H}$  NMR (300 MHz,  $\text{CD}_3\text{OD}$ , complex mixture of rotamers)  $\delta$  7.74-7.82 (m, 2H), 7.36-7.59 (m, 4H), 6.81 (t,  $J = 75$  Hz, 1H), 5.48-5.55 and 4.80-4.89 (m, 1H), 4.71-3.39 (m, 8H), 2.90-2.10 (m, 4H), 1.08 (m, 2H), 0.10 (s, 9H).

25 APCI-MS( $m/z$ ) 701 ( $M+1$ )<sup>+</sup>

(ix) (R)-6-Chloro-8-difluoromethoxy-4-hydroxy-1,1-dioxothiochromane-4-yl-C(O)-Aze-Pab x HOAc

Trifluoroacetic acid (3.0mL) was added to a stirred ice/water-cooled

solution of (*R*)-6-chloro-8-difluoromethoxy-4-hydroxy-1,1-dioxothiochromane-4-yl-Aze-Pab(Teoc) (71 mg; 0.10 mmol; from step (viii) above) in methylene chloride (1 mL). After 1h methylene chloride (3 mL) was added and the solvents were carefully removed under reduced pressure.

- 5 The residue was dissolved in water/acetonitrile and freeze dried. Purification with reversed-phase HPLC (0.1M aq. ammonium acetate/acetonitrile, 7:3) afforded the title compound as a white solid.

Yield: 45 mg (72%).

APCI-MS(*m/z*) 557 (*M*+1)<sup>+</sup>, 555 (*M*-1)<sup>-</sup>

- 10 <sup>1</sup>H NMR (400 MHz; CD<sub>3</sub>OD): δ 7.77 (complex due to diastereomers/rotamers) (d, 0.8H, rotamer); 7.68 (d, 1.2H, rotamer); 7.58 (d, 0.6H, rotamer); 7.55 (d, 0.8H, rotamer); 7.48 (d, 1.2H, rotamer); 7.47 (d, 0.4H, rotamer); 7.40 (m, 0.4H, rotamer); 7.38 (m, 0.6H, rotamer); 6.90 (t, 1H); 5.54 (dd, 0.4H, rotamer); 4.68 (m, 0.6H, rotamer); 4.35-4.61 (several peaks, 3H); 4.09 (m, 0.4H, rotamer); 3.99 (m, 0.6H, rotamer); 3.82 (apparent dt, 0.4H, rotamer); 3.65 (apparent t, 1.2H, rotamer); 3.44 (m, 0.4H, rotamer); 2.69-2.90 (several peaks, 1.4H); 2.48-2.64 (several peaks, 1.6H); 2.29 (m, 0.6H, rotamer); 2.15 (m, 0.4H, rotamer); 1.89 (s, 3H).
- 15
- 20 <sup>13</sup>C NMR (75 MHz; CD<sub>3</sub>OD): (complex due to diastereomers/rotamers, carbonyl and/or amidine carbons) δ 174.1; 173.4; 173.3; 172.9; 168.1.

Example 8

6-Chloro-4-(R)-hydroxy-8-methoxy-1-(R or S)-oxo-thiochromane-4-yl-  
C(O)-Aze-Pab x TFA

- 5 (i) 6-Chloro-4-(R)-hydroxy-8-methoxy-1-(R or S)-oxo-thiochromane-4-  
yl-carboxylic acid (I) and 6-Chloro-4-(R)-hydroxy-8-methoxy-1-(S or R)-  
oxo-thiochromane-4-yl-carboxylic acid (II)

A mixture of (R)-6-chloro-4-hydroxy-8-methoxythiochromane-4-yl-carboxylic acid (700 mg, 2.60 mmol, see Example 1(vi) above) and 35%  
10 H<sub>2</sub>O<sub>2</sub> (3.4 mL, 30.1 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (40 mL) was stirred at 0°C for 2 h, and then overnight at 25°C under nitrogen. The mixture was concentrated *in vacuo* and chromatographed on silica gel eluting with CHCl<sub>3</sub>:MeOH:concentrated NH<sub>4</sub>OH (7.5:2.0:0.5) to afford 320 mg of the ammonium salt of the sub-title compound diastereomer (I) and 190 mg of  
15 the ammonium salt of the sub-title compound diastereomer (II). Both materials were each separately dissolved in H<sub>2</sub>O (40 mL), acidified with 1N HCl, and extracted with EtOAc (2 x 50 mL). The organic extracts of each were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered, and concentrated to afford 290 mg (40%) of sub-title compound (I) and 170 mg (22%) of sub-title compound  
20 (II), respectively.

For sub-title compound diastereomer (I):

<sup>1</sup>H NMR (300 MHz, CD<sub>3</sub>OD) δ 7.25 (s, 1H), 7.05 (s, 1H), 4.00 (s, 3H), 3.60-3.68 (m, 1H), 3.20-3.28 (m, 1H), 2.80-2.92 (m, 1H), 2.46-2.58 (m, 1H)

25 APCI-MS: (M + 1) = 291 m/z

For sub-title compound diastereomer (II):

<sup>1</sup>H NMR (300 MHz, CD<sub>3</sub>OD) δ 7.35 (s, 1H), 7.20 (s, 1H), 4.00 (s, 3H), 3.42-3.58 (m, 1H), 3.18-3.30 (m, 1H), 2.70-2.88 (m, 1H), 2.30-2.45 (m, 1H).

APCI-MS: (M + 1) = 291 m/z

(ii) 6-Chloro-4-(R)-hydroxy-8-methoxy-1-(R or S)-oxo-thiochromane-4-yl-C(O)-Aze-Pab(Teoc)

- 5 To a solution of 6-chloro-4-(R)-hydroxy-8-methoxy-1-(R or S)-oxo-thiochromane-4-yl-carboxylic acid (diastereomer (I) from step (i) above, 284 mg, 0.98 mmol) in DMF (10 mL) under nitrogen was added H-Aze-Pab(Teoc) x HCl (571 mg, 1.27 mmol), PyBOP (559 mg, 1.07 mmol), and DIPEA (315 mg, 2.44 mmol) at 0°C. The reaction was stirred at 0°C  
10 for 2 h and then at room temperature overnight. The mixture was concentrated *in vacuo* and the residue chromatographed twice on silica gel, eluting twice with CHCl<sub>3</sub>:EtOH (9:1) to afford 190 mg (30%) of the sub-title compound as a white solid.

- 15 mp: 225-230°C

R<sub>f</sub> = 0.57 (9:1 CHCl<sub>3</sub>:EtOH)

- <sup>1</sup>H NMR (300 MHz, CD<sub>3</sub>OD/CDCl<sub>3</sub> (1:1), complex mixture of rotamers)  
δ 7.83 (d, *J* = 8 Hz, 2H), 7.38 (d, *J* = 7 Hz, 2H), 7.08 (s, 1H), 6.99 (s, 1H), 4.87-4.92 (m, 1H), 4.51-4.54 (m, 2H), 4.21-4.30 (m, 4H), 4.00 (s, 3H), 3.12-3.39 (m, 4H), 2.39-2.50 (m, 1H), 2.16-2.22 (m, 1H), 1.08-1.14 (m, 2H), 0.08 (s, 9H)  
20

APCI-MS: (M + 1) = 649 m/z

- (iii) 6-Chloro-4-(R)-hydroxy-8-methoxy-1-(R or S)-oxo-thiochromane-4-yl-C(O)-Aze-Pab x TFA  
25

To a solution of 6-chloro-4-(R)-hydroxy-8-methoxy-1-(R or S)-oxo-thiochromane-4-yl-C(O)-Aze-Pab(Teoc) (90 mg, 0.14 mmol, from step (ii) above) in dichloromethane (2 mL), trifluoroacetic acid (2 mL) was added. The solution was left at ambient temperature for 1 h whereupon it was

concentrated under reduced pressure. The residue was dissolved in deionized water and freeze dried. Yield: 80 mg (0.13 mmol).

<sup>1</sup>H-NMR (500 MHz; CD<sub>3</sub>OD): (complex mixture due to rotamers) δ 8.75  
5 (t, 0.75H, rotamer); 8.63 (t, 0.25H, rotamer); 7.76 (d, 0.5H, rotamer);  
7.72 (d, 1.5H, rotamer); 7.55 (d, 2H, mix of rotamers); 7.23 (s, 1H, mix  
of rotamers); 7.11 (s, 0.75H, rotamer); 7.01 (s, 0.25H, rotamers); 5.45  
(m, 0.5H); 4.61 (m, 1H); 4.47 (m, 2H); 4.13 (dm, 0.5H); 3.98 (s, 3H);  
3.83 (q, 1H); 3.23 (m, 1.5H); 3.10 (m, 1.25H); 2.97 (m, 0.25H); 2.78  
10 (m, 0.25H); 2.49 (m, 0.75H); 2.26 (m, 1H); 2.18 (m, 1H)  
MS (m/z) 505 (M+1)<sup>+</sup>, 503 (M-1)<sup>-</sup>

#### Example 9

6-Chloro-4-(R)-hydroxy-8-methoxy-1-(S or R)-oxo-thiochromane-4-yl-  
15 C(O)-Aze-Pab x TFA

(i) 6-Chloro-4-(R)-hydroxy-8-methoxy-1-(S or R)-oxo-thiochromane-4-  
yl-C(O)-Aze-Pab(Teoc)

To a solution of 6-chloro-4-(R)-hydroxy-8-methoxy-1-(S or R)-oxo-  
20 thiochromane-4-yl-carboxylic acid (diastereomer (II) from Example 8(i)  
above, 166 mg, 0.57 mmol) in DMF (10 mL) under nitrogen were added  
H-Aze-Pab(Teoc) x HCl (333 mg, 0.74 mmol), PyBOP (327 mg, 0.63  
mmol), and DIPEA (184 mg, 1.43 mmol) at 0°C. The reaction was  
stirred at 0°C for 2 h and then at room temperature overnight. The  
25 mixture was concentrated *in vacuo* and the residue chromatographed twice  
on silica gel, eluting first with CHCl<sub>3</sub>:EtOH (9:1) and second with  
CHCl<sub>3</sub>:MeOH (20:1) to afford 80 mg (22%) of the sub-title compound as  
a white solid.



mp: 140-145°C

R<sub>f</sub> = 0.60 (9:1 CHCl<sub>3</sub>:EtOH)

<sup>1</sup>H NMR (300 MHz, CD<sub>3</sub>OD, complex mixture of rotamers) δ 7.80 and  
7.72 (d, *J* = 8 Hz, 2H), 7.42-7.49 (m, 2H), 7.32 (s, 1H), 7.17 (s, 1H),  
5 5.52-5.57 (m, 1H), 4.44-4.91 (m, 4H), 4.19-4.25 (m, 2H), 3.97 (s, 3H),  
3.00-3.72 (m, 3H), 2.10-2.82 (m, 3H), 1.04-1.11 (m, 2H), 0.07 (s, 9H)

APCI-MS: (M + 1) = 649 m/z

(ii) 6-Chloro-(R)-4-(R)-hydroxy-8-methoxy-1-(S or R)-oxo-thiochrom-  
10 ane-4-yl-C(O)-Aze-Pab x TFA

To a solution of 6-chloro-4-(R)-hydroxy-8-methoxy-1-(S or R)-oxo-  
thiochromane-4-yl-C(O)-Aze-Pab(Teoc) (35 mg, 0.054 mmol from step (i)  
above) in dichloromethane (2 mL), was added trifluoroacetic acid (2 mL).  
The solution was left at ambient temperature for 1 h whereupon it was  
15 concentrated at reduced pressure. The residue was dissolved in deionized  
water and freeze dried. Yield: 30 mg (0.048 mmol).

<sup>1</sup>H-NMR (500 MHz; CD<sub>3</sub>OD): (complex mixture due to rotamers) δ 8.82  
(m, 0.2H); 8.65 (m, 0.2H); 7.77 (d, 0.9H); 7.66 (d, 0.8H); 7.59 (d,  
20 0.9H); 7.43 (d, 0.8H); 7.40 (d, 0.4H); 7.30 (d, 0.4H); 7.16 (dd, 0.8H);  
5.55 (m, 0.45H); 4.70 (m, 0.6H); 4.59 (m, 0.6H); 4.51 (m, 1.8H); 4.05  
(q, 0.45H); 3.97 (s, 3H); 3.94 (m, 0.4H); 3.58 (m, 0.4H); 3.38 (m,  
0.4H); 3.16 (m, 0.4H); 3.03 (m, 0.4H); 2.75 (m, 0.8H); 2.58 (m, 0.4H);  
2.45 (m, 1.2H); 2.30 (m, 0.4H); 2.15 (m, 0.45H)

25 MS (m/z): 505 (M+1)<sup>+</sup>, 503 (M-1)<sup>-</sup>

Example 10

Title compounds of Examples 1, 2, 3, 7, 8 and 9 were tested in Test A  
above and were found to exhibit an IC<sub>50</sub>TT value of less than 0.05 μM.

Example 11

Title compounds of Examples 4, 5 and 6 are tested in Test E above and are found to exhibit oral and/or parenteral bioavailability in the rat as the  
5 corresponding active inhibitor (free amidine).

Example 12

Title compounds of Examples 4, 5 and 6 are tested in Test G above and are found to exhibit oral and/or parenteral bioavailability in the rat as the  
10 corresponding active inhibitor (free amidine).

**Abbreviations**

	Ac	=	acetyl
15	AcOH	=	acetic acid
	API	=	atmospheric pressure ionisation (in relation to MS)
	aq.	=	aqueous
	AUC	=	area under the curve
	Aze	=	(S)-azetidine-2-carboxylate (unless otherwise specified)
20	AzeOH	=	azetidine-2-carboxylic acid
	Bn	=	benzyl
	BSA	=	bovine serum albumin
	CI	=	chemical ionisation (in relation to MS)
	d	=	day(s)
25	DCC	=	dicyclohexyl carbodiimide
	DIPEA	=	diisopropylethylamine
	DMAP	=	4-(N,N-dimethyl amino) pyridine
	DMF	=	dimethylformamide
	DMSO	=	dimethylsulfoxide

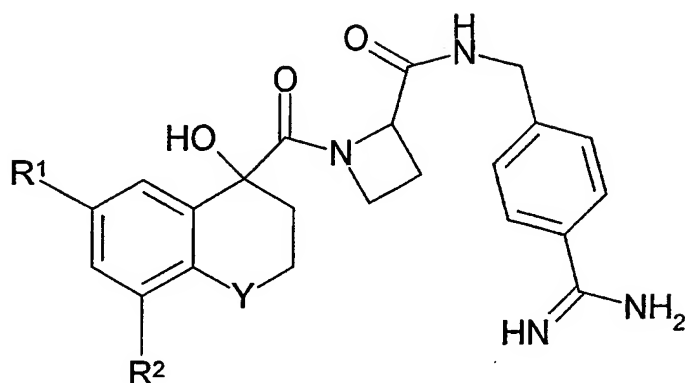
	EDC	=	1-(3-dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride
	Et	=	ethyl
	ether	=	diethyl ether
5	EtOAc	=	ethyl acetate
	EtOH	=	ethanol
	h	=	hour(s)
	HATU	=	<i>O</i> -(azabenzotriazol-1-yl)- <i>N,N,N',N'</i> -tetramethyluronium hexafluorophosphate
10	HBTU	=	[ <i>N,N,N',N'</i> -tetramethyl- <i>O</i> -(benzotriazol-1-yl)uronium hexafluorophosphate]
	HCl	=	hydrochloric acid, hydrogen chloride gas or hydrochloride salt (depending on context)
	Hex	=	hexanes
15	HOAc	=	acetic acid
	HPLC	=	high performance liquid chromatography
	LC	=	liquid chromatography
	Me	=	methyl
	MeOH	=	methanol
20	min.	=	minutes
	MS	=	mass spectroscopy
	MTBE	=	methyl <i>tert</i> -butyl ether
	NADH	=	nicotinamide adenine dinucleotide, reduced form
	NADPH	=	nicotinamide adenine dinucleotide phosphate, reduced form
25	NIH	=	National Institute of Health (US)
	NIHU	=	National Institute of Health units
	Pab	=	<i>para</i> -amidinobenzylamino
	H-Pab	=	<i>para</i> -amidinobenzylamine

	PTSA	=	<i>para</i> -toluenesulphonic acid
	PyBOP	=	(benzotriazol-1-yloxy)tripyrrolidinophosphonium hexafluorophosphate
	RPLC	=	reverse phase high performance liquid chromatography
5	rt	=	room temperature
	SOPs	=	standard operating procedures
	TBTU	=	[ <i>N,N,N',N'</i> -tetramethyl- <i>O</i> -(benzotriazol-1-yl)uronium tetrafluoroborate]
	TEA	=	triethylamine
10	Teoc	=	2-(trimethylsilyl)ethoxycarbonyl
	TFA	=	trifluoroacetic acid
	THF	=	tetrahydrofuran
	THP	=	tetrahydropyranyl
	TLC	=	thin layer chromatography
15	TMSCN	=	trimethylsilyl cyanide
	Z	=	benzyloxycarbonyl

Prefixes *n*, *s*, *i* and *t* have their usual meanings: normal, secondary, iso and tertiary.

# Claims

1. A compound of formula I,



5

wherein

Y represents S(O) or S(O)<sub>2</sub>;

R<sup>1</sup> represents halo; and

- 10 R<sup>2</sup> represents H, halo or C<sub>1-4</sub> alkoxy (which latter group is optionally substituted by one or more halo groups);  
or a pharmaceutically acceptable derivative thereof.

2. A compound as claimed in Claim 1 wherein R<sup>1</sup> represents chloro.

15

3. A compound as claimed in Claim 1 or Claim 2, wherein R<sup>2</sup> represents H, halo or C<sub>1-2</sub> alkoxy (which latter group is optionally substituted by one or more halo groups).

- 20 4. A compound as claimed in any one of Claims 1 to 3, wherein R<sup>1</sup> represents chloro and R<sup>2</sup> represents H, chloro, OCH<sub>3</sub>, OCHF<sub>2</sub> or OCF<sub>3</sub>.

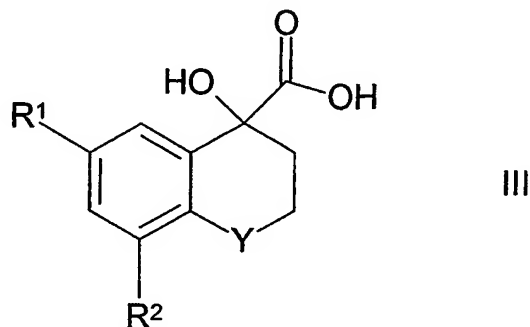
5. A compound as claimed in Claim 4, wherein  $R^2$  represents  $OCH_3$ .

6. (*R*)-6-Chloro-4-hydroxy-8-methoxy-1,1-dioxothiochromane-4-yl-C(O)-(*S*)-Aze-Pab or a pharmaceutically acceptable derivative thereof.

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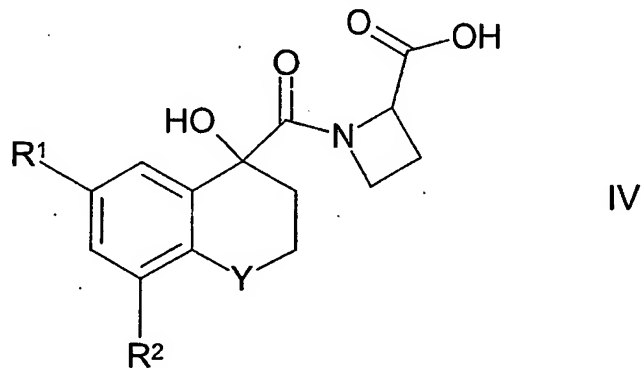
7. A process for the preparation of a compound of formula I as defined in Claim 1, which comprises:

(i) the coupling of a compound of formula III,



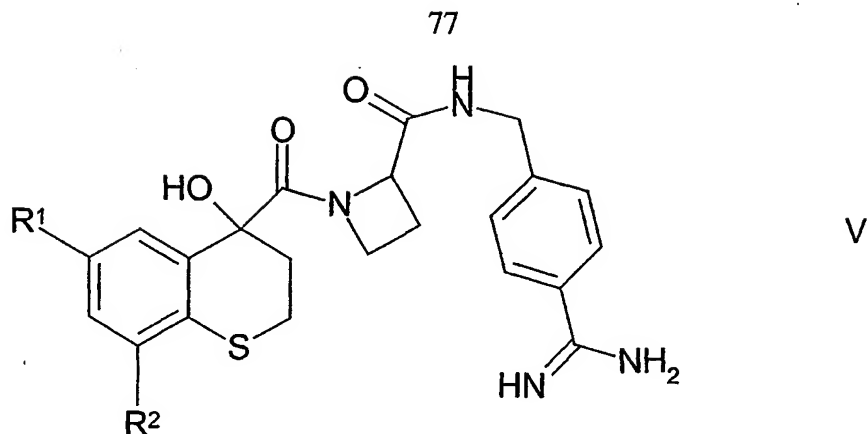
10 wherein Y,  $R^1$  and  $R^2$  are as defined in Claim 1, with 4-amidinobenzyl-2-azetidinecarboxamide;

(ii) the coupling of a compound of formula IV,



15 wherein Y,  $R^1$  and  $R^2$  are as defined in Claim 1, with *para*-amidino-benzylamine;

(iii) complete or partial oxidation of a corresponding compound of formula V,

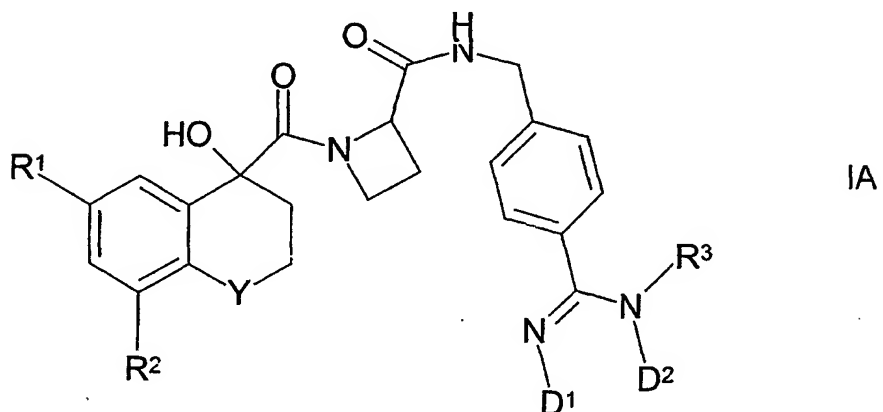


wherein  $R^1$  and  $R^2$  are as defined in Claim 1;

(iv) deprotection of a protected derivative of a compound of formula I as defined in Claim 1; or

- 5 (v) introduction or interconversion of a substituent on the aromatic ring in a compound of formula I as defined in Claim 1.

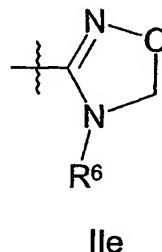
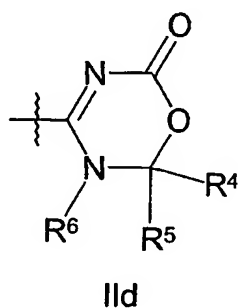
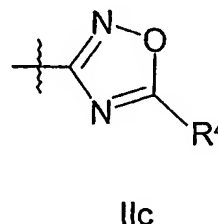
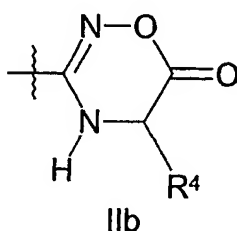
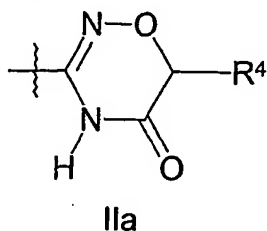
8. A compound as claimed in Claim 1 which is a pharmaceutically acceptable derivative of a compound of formula I, which derivative is a  
10 compound of formula IA,



wherein

Y,  $R^1$  and  $R^2$  are as defined in any one of Claims 1 to 6;

D<sup>1</sup> and D<sup>2</sup> independently represent H, -OR<sup>7</sup> or R<sup>8</sup>, or D<sup>1</sup>, D<sup>2</sup> and R<sup>3</sup> together with the amidine group to which they are attached, form a cyclic group of formula IIa, IIb, IIc, IId or IId, or IId,



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wherein wavy lines indicate the points of attachment to the benzene ring;

R<sup>3</sup> represents H, or R<sup>3</sup>, D<sup>1</sup>, and D<sup>2</sup> together with the amidine group to which they are attached, form a cyclic group of formula IIa, IIb, IIc, IId or IId;

10 R<sup>4</sup> and R<sup>5</sup> independently represent H or C<sub>1-4</sub> alkyl;

R<sup>6</sup> represents H, C<sub>1-6</sub> alkyl (which latter group is optionally substituted by one or more halo groups) or C(O)OR<sup>12</sup>;

R<sup>7</sup> represents H, C<sub>6-10</sub> aryl, C<sub>1-10</sub> alkyl (which latter group is optionally substituted by one or more halo groups), C<sub>1-3</sub> alkylphenyl, -C(R<sup>9a</sup>)(R<sup>9b</sup>)R<sup>10</sup>,

15 -C(O)R<sup>11a</sup>, -C(O)OR<sup>12</sup>, -C(O)N(R<sup>13</sup>)R<sup>14</sup> or -(CH<sub>2</sub>)<sub>n</sub>(O)<sub>m</sub>R<sup>15</sup>;

R<sup>8</sup> represents -C(R<sup>9a</sup>)(R<sup>9b</sup>)R<sup>10</sup>, -C(O)R<sup>11b</sup> or -C(O)OR<sup>12</sup>;

R<sup>9a</sup> and R<sup>9b</sup> independently represent, at each occurrence, H or C<sub>1-6</sub> alkyl;

R<sup>10</sup> represents, at each occurrence, -OC(O)R<sup>16a</sup>, -OC(O)OR<sup>17</sup>,

-N(R<sup>18a</sup>)C(O)OR<sup>17</sup> or -OC(O)N(R<sup>18b</sup>)R<sup>17</sup>;



$R^{11a}$  and  $R^{11b}$  independently represent, at each occurrence,  $C_{6-10}$  aryl,  $C_{1-3}$  alkylphenyl (which latter two groups are optionally substituted by one or more substituents selected from  $C_{1-6}$  alkyl and halo),  $-[C(R^{19a})(R^{19b})]_pOC(O)R^{16b}$ , or  $R^{11a}$  represents  $C_{1-17}$  alkyl (optionally substituted by  $C_{1-6}$  alkoxy,  $C_{1-6}$  acyloxy, amino or halo) or  $R^{11b}$  represents  $C_{1-6}$  alkyl;

$R^{12}$  represents, at each occurrence,  $C_{1-17}$  alkyl (optionally substituted by one or more substituents selected from  $C_{1-6}$  alkoxy,  $C_{1-6}$  acyloxy,  $-Si(R^{20a})(R^{20b})(R^{20c})$  and halo),  $C_{6-10}$  aryl,  $C_{1-3}$  alkylphenyl (which latter two groups are optionally substituted by  $C_{1-6}$  alkyl,  $C_{1-6}$  alkoxy, cyano or halo),  $-[C(R^{19a})(R^{19b})]_qOC(O)R^{16b}$  or  $-CH_2R^{21}$ ;

$R^{13}$  represents H or  $C_{1-7}$  alkyl, or together with  $R^{14}$  represents  $C_{4-5}$  alkylene;

$R^{14}$  represents  $C_{6-10}$  aryl or  $C_{1-10}$  alkyl (which latter group is optionally substituted by one or more substituents selected from OH, halo,  $CO_2H$ ,  $C_{1-6}$  alkoxy,  $C_{1-6}$  acyloxy and  $C_{6-10}$  aryl), or together with  $R^{13}$  represents  $C_{4-5}$  alkylene;

$R^{15}$  represents  $C_{1-7}$  alkyl optionally substituted by one or more  $-OC(O)C(H)(R^{22})N(G)(G^a)$  groups;

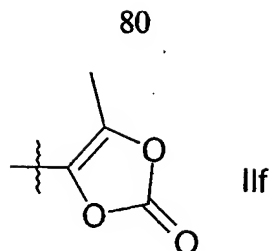
$R^{16a}$ ,  $R^{16b}$  and  $R^{17}$  independently represent, at each occurrence,  $C_{6-10}$  aryl or  $C_{1-17}$  alkyl (which latter group is optionally substituted by one or more substituents selected from  $-OH$ , halo,  $-CO_2H$ ,  $C_{1-6}$  alkoxy,  $C_{1-6}$  acyloxy and  $C_{6-10}$  aryl), or  $R^{16b}$  represents  $C_{1-6}$  alkoxy (optionally substituted by one or more substituents selected from  $C_{1-6}$  alkyl and halo);

$R^{18a}$  and  $R^{18b}$  independently represent H or  $C_{1-4}$  alkyl;

$R^{19a}$  and  $R^{19b}$  independently represent, at each occurrence, H or  $C_{1-6}$  alkyl;

$R^{20a}$  to  $R^{20c}$  independently represent, at each occurrence,  $C_{1-6}$  alkyl or phenyl;

$R^{21}$  represents the structural fragment II f



$R^{22}$  represents  $C_{3-4}$  alkyl;

G and  $G^a$  independently represent H, an amino protective group, or G and

$G^a$  together represent an amino protective group;

5 m represents 0 or 1;

n represents 1, 2 or 3;

p represents 3 or 4;

q represents 2 or 3;

or a pharmaceutically acceptable salt thereof,

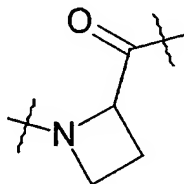
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provided that:

(a)  $D^1$  and  $D^2$  do not both represent H; and

(b) when one of  $D^1$  and  $D^2$  represents  $-OR^7$ , then the other represents H.

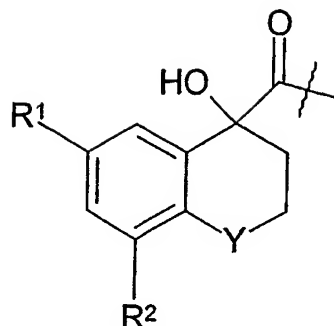
15 9. A compound as defined in any one of Claims 1 to 5 or 8, wherein the fragment



is in the *S*-configuration.

20 10. A compound of formula I as defined in any one of Claims 1 to 5 or 8 or 9, wherein the fragment

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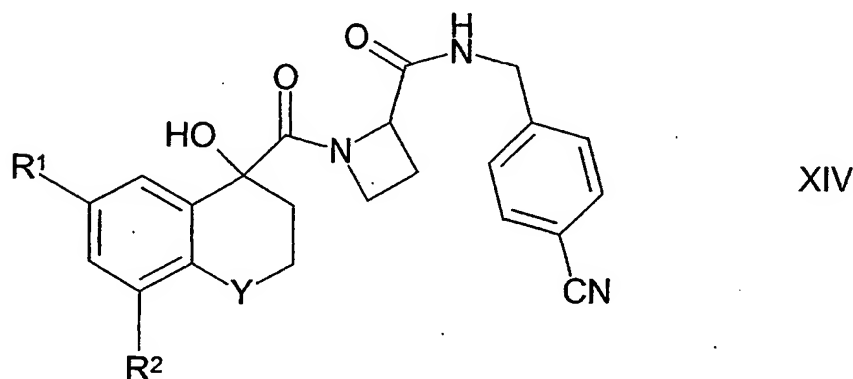


is in the *R*-configuration.

11. A process for the preparation of a compound of formula IA as  
5 defined in Claim 8, which comprises:

(1) for compounds of formula IA in which one of D<sup>1</sup> and D<sup>2</sup> represents  
-OR<sup>7</sup> (in which R<sup>7</sup> represents H, C<sub>6-10</sub> aryl, C<sub>1-10</sub> alkyl (which latter group  
is optionally substituted by one or more halo groups) or C<sub>1-3</sub> alkylphenyl),  
reaction of a compound of formula XIV,

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XIV

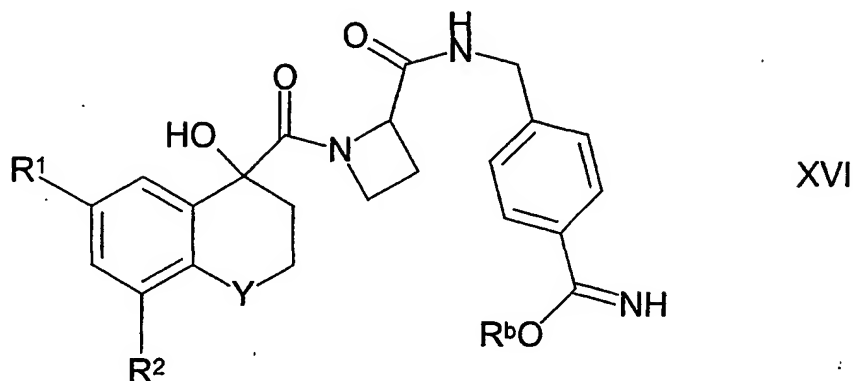
wherein Y, R<sup>1</sup> and R<sup>2</sup> are as defined in Claim 8, with a compound of  
formula XV,



XV

15 wherein R<sup>a</sup> represents H, C<sub>6-10</sub> aryl, C<sub>1-10</sub> alkyl (which latter group is  
optionally substituted by one or more halo groups) or C<sub>1-3</sub> alkylphenyl,  
optionally by pre-treating the compound of formula XIV with gaseous

HCl, in the presence of a lower alkyl alcohol to form a compound of formula XVI,



wherein  $R^b$  represents lower alkyl and Y,  $R^1$  and  $R^2$  are as defined in Claim 8;

(2) for compounds of formula IA in which one of  $D^1$  and  $D^2$  represents  $-OR^7$  (in which  $R^7$  represents H,  $C_{6-10}$  aryl,  $C_{1-10}$  alkyl (which latter group is optionally substituted by one or more halo groups) or  $C_{1-3}$  alkylphenyl), reaction of a corresponding compound of formula IA, in which one of  $D^1$  and  $D^2$  represents  $-C(O)OR^{12}$  and the other represents H, with a compound of formula XV, as defined above, followed by removal of the  $-C(O)OR^{12}$  group;

(3) for compounds of formula IA in which  $D^1$  or  $D^2$  represents  $R^8$ , reaction of a corresponding compound of formula I, or a corresponding compound of formula IA in which  $D^1$  or  $D^2$  (as appropriate) represents H, with a compound of formula XVII,



wherein  $L^3$  represents a leaving group and  $R^8$  is as defined in Claim 8;

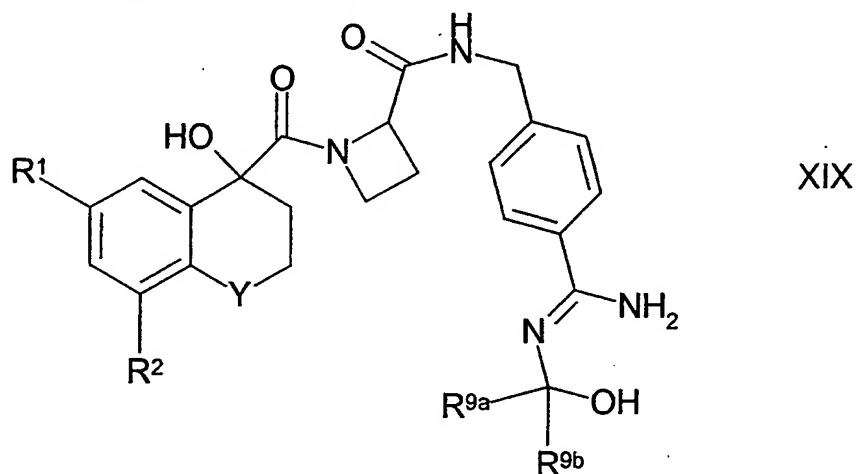
(4) for compounds of formula IA in which one of  $D^1$  and  $D^2$  represents  $-OR^7$  (wherein  $R^7$  does not represent H), reaction of a corresponding compound of formula IA in which one of  $D^1$  and  $D^2$  represents  $-OH$  with a compound of formula XVIII,



XVIII

wherein  $R^{7a}$  represents  $R^7$  as defined in Claim 8, except that it does not represent H, and  $L^3$  is as defined above;

- (5) for compounds of formula IA in which one of  $D^1$  and  $D^2$  represents H and the other represents  $-C(R^{9a})(R^{9b})R^{10}$ , wherein  $R^{10}$  represents  $-OC(O)R^{16a}$ ,  $-OC(O)OR^{17}$  or  $-OC(O)N(R^{18b})R^{17}$ , reaction of a corresponding compound of formula XIX,



- wherein Y,  $R^1$ ,  $R^2$ ,  $R^{9a}$  and  $R^{9b}$  are as defined in Claim 8, with a compound of formula XX,



XX

wherein  $R^c$  represents  $R^{16a}$ ,  $-OR^{17}$  or  $-N(R^{18b})R^{17}$ , and  $R^{16a}$ ,  $R^{17}$  and  $R^{18b}$  are as defined in Claim 8 and  $L^3$  is as defined above;

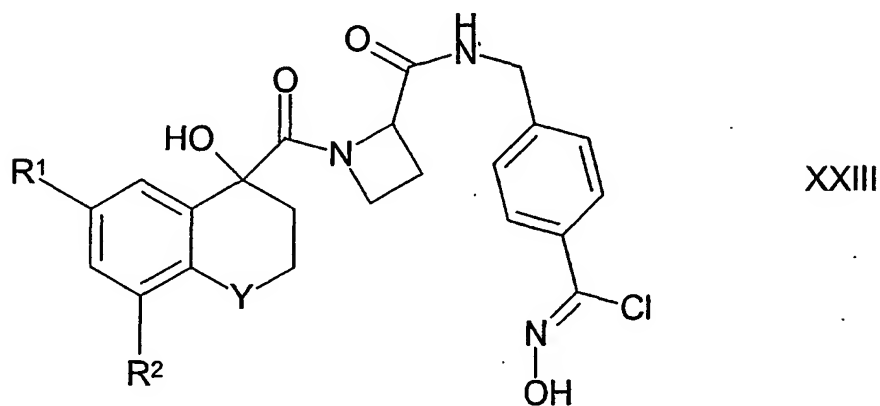
- (6) for compounds of formula IA in which  $D^1$ ,  $D^2$  and  $R^3$ , together with the amidine group to which they are attached, represent a group IIa as defined in Claim 8, reaction of a corresponding compound of formula IA in which  $D^1$  represents OH with a compound of formula XXII,



XXII

- wherein Hal represents halo and  $R^4$  is as defined in Claim 8, followed by cyclisation of the resultant intermediate;

(7) for compounds of formula IA in which D<sup>1</sup>, D<sup>2</sup> and R<sup>3</sup>, together with the amidine group to which they are attached, represent a group IIb as defined in Claim 8, reaction of a corresponding compound of formula XXIII,



wherein Y, R<sup>1</sup> and R<sup>2</sup> are as defined in Claim 8, with a compound of formula XXIV,



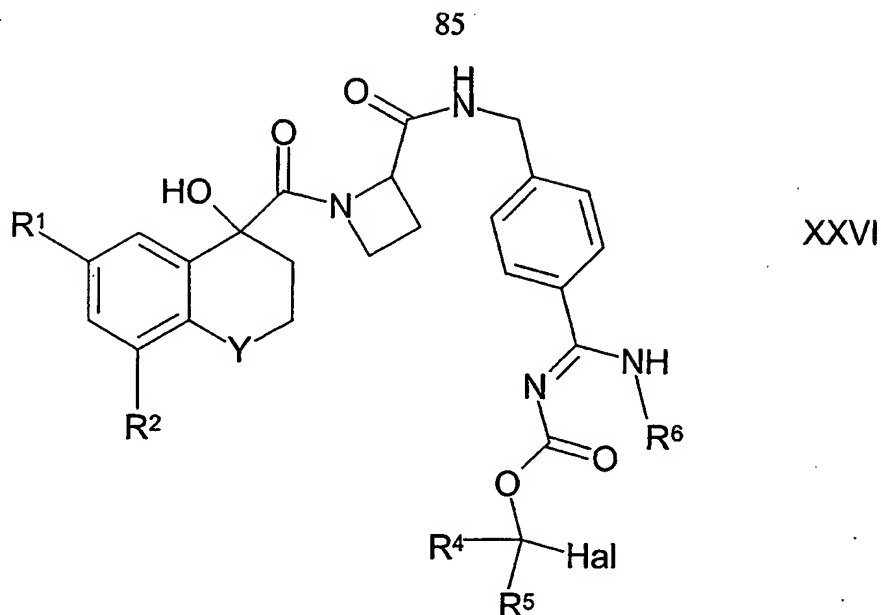
wherein R<sup>b</sup> is as defined above and R<sup>4</sup> is as defined in Claim 8;

(8) for compounds of formula IA in which D<sup>1</sup>, D<sup>2</sup> and R<sup>3</sup>, together with the amidine group to which they are attached, represent a group IIc as defined in Claim 8, reaction of a corresponding compound of formula IA in which D<sup>1</sup> represents OH with a compound of formula XXV,



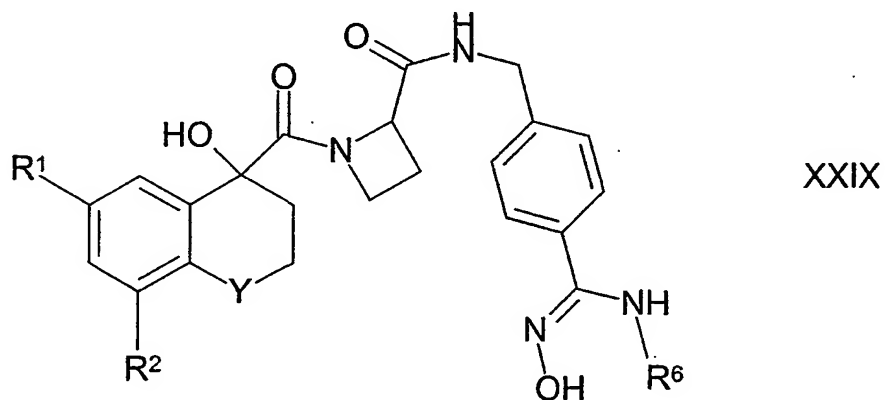
wherein R<sup>4</sup> is as defined in Claim 8;

(9) for compounds of formula IA in which D<sup>1</sup>, D<sup>2</sup> and R<sup>3</sup>, together with the amidine group to which they are attached, represent a group IId as defined in Claim 8, cyclisation of a compound of formula XXVI,



wherein Hal is as defined above and Y, R<sup>1</sup>, R<sup>2</sup>, R<sup>4</sup>, R<sup>5</sup> and R<sup>6</sup> are as defined in Claim 8;

- (10) for compounds of formula IA in which D<sup>1</sup>, D<sup>2</sup> and R<sup>3</sup>, together with the amidine group to which they are attached, represent a group Ile as defined in Claim 8, reaction of a corresponding compound of formula XXIX,



wherein Y, R<sup>1</sup>, R<sup>2</sup> and R<sup>6</sup> are as defined in Claim 8 with formaldehyde;

- (11) deprotection of a protected derivative of a compound of formula IA as defined in Claim 8; or

(12) introduction or interconversion of a substituent on an aromatic and/or non-aromatic, carbocyclic and heterocyclic ring(s) in compounds of formula IA as defined in Claim 8.

- 5 12. A pharmaceutical formulation including a compound as defined in any one of Claims 1 to 6 or 8 to 10, or a pharmaceutically acceptable derivative or salt thereof (as appropriate), in admixture with a pharmaceutically acceptable adjuvant, diluent or carrier.
- 10 13. A compound as defined in any one of Claims 1 to 6 or 8 to 10, or a pharmaceutically acceptable derivative or salt thereof (as appropriate), for use as a pharmaceutical.
14. A compound as defined in any one of Claims 1 to 6 or 8 to 10, or a  
15 pharmaceutically acceptable derivative or salt thereof (as appropriate), for use in the treatment of a condition where inhibition of thrombin is required.
15. A compound as defined in any one of Claims 1 to 6 or 8 to 10, or a  
20 pharmaceutically acceptable derivative or salt thereof (as appropriate), for use in the treatment of thrombosis.
16. A compound as defined in any one of Claims 1 to 6 or 8 to 10, or a  
25 pharmaceutically acceptable derivative or salt thereof (as appropriate), for use as an anticoagulant.
17. The use of a compound as defined in any one of Claims 1 to 6 or 8 to 10, or a pharmaceutically acceptable derivative or salt thereof (as



appropriate), as active ingredient in the manufacture of a medicament for the treatment of a condition where inhibition of thrombin is required.

18. The use as claimed in Claim 17, wherein the condition is thrombosis.

5

19. The use as claimed in Claim 17, wherein the condition is hypercoagulability in blood and/or tissues.

20. The use of a compound as defined in any one of Claims 1 to 6 or 8  
10 to 10, or a pharmaceutically acceptable derivative or salt thereof (as appropriate), as active ingredient in the manufacture of an anticoagulant.

21. A method of treatment of a condition where inhibition of thrombin is required which method comprises administration of a therapeutically  
15 effective amount of a compound as defined in any one of Claims 1 to 6 or 8 to 10, or a pharmaceutically acceptable derivative or salt thereof (as appropriate), to a person suffering from, or susceptible to, such a condition.

20 22. A method as claimed in Claim 21, wherein the condition is thrombosis.

23. A method as claimed in Claim 21, wherein the condition is hypercoagulability in blood and/or tissues.

## INTERNATIONAL SEARCH REPORT

International application No.

PCT/SE 01/01052

## A. CLASSIFICATION OF SUBJECT MATTER

IPC7: C07D 409/06, A61K 31/397, A61K 31/382, A61P 7/00 // (C07D 409/06, 335:06, 205:04)

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC7: C07D

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

SE,DK,FI,NO classes as above

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

EPO-INTERNAL, WPI DATA, PAJ, CHEM.ABS DATA

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	WO 9857932 A1 (ASTRA AKTIEBOLAG), 23 December 1998 (23.12.98), page 75, line 17 - page 89, line 7, the claims	1-23
	--	
P,A	WO 0035869 A1 (ASTRAZENECA AB), 22 June 2000 (22.06.00)	1-23
	---	
A	WO 9702284 A1 (ASTRA AKTIEBOLAG), 23 January 1997 (23.01.97)	1-23
	-- -----	

☐ Further documents are listed in the continuation of Box C.
 ☒ See patent family annex.

\* Special categories of cited documents:

"A" document defining the general state of the art which is not considered to be of particular relevance

"E" earlier application or patent but published on or after the international filing date

"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)

"O" document referring to an oral disclosure, use, exhibition or other means

"P" document published prior to the international filing date but later than the priority date claimed

"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X" document of particular relevance: the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

"Y" document of particular relevance: the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art

"&amp;" document member of the same patent family

Date of the actual completion of the international search

22 August 2001

Date of mailing of the international search report

28-08-2001

Name and mailing address of the ISA:

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# INTERNATIONAL SEARCH REPORT

International application No.  
**PCT/SE01/01052**

## Box I Observations where certain claims were found unsearchable (Continuation of item 1 of first sheet)

This international search report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:

1. ☒ Claims Nos.: **21-23**  
because they relate to subject matter not required to be searched by this Authority, namely:  
**see next sheet**
2. ☐ Claims Nos.:  
because they relate to parts of the international application that do not comply with the prescribed requirements to such an extent that no meaningful international search can be carried out, specifically:
3. ☐ Claims Nos.:  
because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).

## Box II Observations where unity of invention is lacking (Continuation of item 2 of first sheet)

This International Searching Authority found multiple inventions in this international application, as follows:

1. ☐ As all required additional search fees were timely paid by the applicant, this international search report covers all searchable claims.
2. ☐ As all searchable claims could be searched without effort justifying an additional fee, this Authority did not invite payment of any additional fee.
3. ☐ As only some of the required additional search fees were timely paid by the applicant, this international search report covers only those claims for which fees were paid, specifically claims Nos.:
4. ☐ No required additional search fees were timely paid by the applicant. Consequently, this international search report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:

### Remark on Protest

- ☐ The additional search fees were accompanied by the applicant's protest.  
☐ No protest accompanied the payment of additional search fees.

INTERNATIONAL SEARCH REPORT

International application No.  
PCT/SE01/01052

Claims 21-23 relate to methods of treatment of the human or animal body by surgery or by therapy/ diagnostic methods practised on the human or animal body/Rule 39.1.(iv). Nevertheless, a search has been executed for these claims. The search has been based on the alleged effects of the compounds/compositions.

**INTERNATIONAL SEARCH REPORT**  
Information on patent family members

02/08/01

International application No.

PCT/SE 01/01052

Patent document cited in search report			Publication date	Patent family member(s)		Publication date
WO	9857932	A1	23/12/98	AU	7946198 A	04/01/99
				BR	9810177 A	08/08/00
				CN	1266427 T	13/09/00
				EE	9900600 A	15/08/00
				EP	1012140 A	28/06/00
				NO	996313 A	21/02/00
				PL	337645 A	28/08/00
				SE	9702378 D	00/00/00
				SK	167099 A	12/06/00
				TR	9903134 T	00/00/00
				ZA	9805036 A	12/01/99
				AU	727680 B	21/12/00
				AU	7946098 A	04/01/99
				EP	1009570 A	21/06/00
				HU	0001787 A	28/09/00
				SE	9801099 D	00/00/00
				US	6243956 B	12/06/01
WO	0035869	A1	22/06/00	AU	2018700 A	03/07/00
				SE	9804313 D	00/00/00
WO	9702284	A1	23/01/97	AU	709710 B	02/09/99
				AU	6325396 A	05/02/97
				BR	9609447 A	02/03/99
				CA	2226256 A	23/01/97
				CN	1195354 A	07/10/98
				CZ	9704212 A	17/06/98
				EE	9700354 A	15/06/98
				EP	0836615 A	22/04/98
				GB	9526411 D	00/00/00
				HU	9802315 A	28/01/99
				IL	122813 D	00/00/00
				JP	11508591 T	27/07/99
				NO	980015 A	25/02/98
				PL	324483 A	25/05/98
				SK	180497 A	04/11/98
				TR	9800009 T	00/00/00
				US	6051568 A	18/04/00
				SE	9502487 D	00/00/00
				ZA	9605344 A	06/01/97
				SE	9502504 D	00/00/00
				SE	9502505 D	00/00/00
				SE	9503923 D	00/00/00
				SE	9504349 D	00/00/00